Crystallization of a Photosensitive Nitrile Hydratase from *Rhodococcus* sp. N-771

Teruyuki Nagamune^{1,2}[†], Jun Honda¹, Won-Dai Cho²[‡], Nobuo Kamiya³, Yoshitaka Teratani⁴, Akira Hirata⁴, Hiroyuki Sasabe¹ and Isao Endo²

¹Bioelectronic Materials Laboratory, Frontier Research Program
 ²Chemical Engineering Laboratory
 ³X-Ray Crystallography Laboratory
 The Institute of Physical and Chemical Research (RIKEN),
 Wako-Shi, Saitama 351-01, Japan

 ⁴Department of Applied Chemistry, School of Science and Engineering
 Waseda University, Shinjuku-ku, Tokyo 169, Japan

(Received 5 March 1991; accepted 15 March 1991)

A photosensitive nitrile hydratase from *Rhodococcus* sp. N-771 has been crystallized in two different crystal forms in its inactive form. One crystal form belongs to an orthorhombic space group $P2_12_12$ with unit cell dimensions of a = 117.4 Å, b = 145.7 Å and c = 52.1 Å, and the other form belongs to a hexagonal space group $P6_322$ with unit cell dimensions of a = 110.2 Å and c = 412.1 Å.

Keywords: crystallization; nitrile hydratase: Rhodococcus: photosensitive: crystallographic parameter

Nitrile hydratase (NHase§) hydrates various nitrile compounds to the corresponding amide groups. The NHase enzymes from Rhodococcus sp. N-771 and sp. N-774 show an intriguing property towards light, and have both active and inactive forms. We have already reported that the NHase can be inactivated in the dark in vivo depending on the conditions of both dissolved oxygen concentration and temperature, and that the inactivated NHase can be reactivated by light (Nagamune et al., 1990a). In addition, the inactive enzyme purified from the inactivated cells could also be photoactivated in vitro by near-ultra-violet light irradiation, but the active enzyme could not be inactivated by dark and aerobic incubation (Nagamune et al., 1990b). Upon photo-irradiation, drastic changes of NHase spectra are observed in absorption, circular dichroism, electron spin resonance and Mössbauer measurements, and these changes are most probably due to the electronic changes of the iron center composed of prosthetic

non-heme iron and PQQ or a PQQ-like compound (unpublished results).

The NHase gene of *Rhodococcus* sp. N-774 has been cloned, and the amino acid sequence of the enzyme deduced from the nucleotide sequence of the cloned gene indicated that the NHase consists of two heterogeneous subunits, α (207 amino acid residues, relative molecular mass $M_r = 22,918$) and β (212 amino acid residues, relative molecular mass $M_r = 23,428$; Ikehata *et al.*, 1989). Recently, we also cloned the NHase gene from *Rhodococcus* sp. N-771 and clarified that the nucleotide sequence of the gene was identical with that from *Rhodococcus* sp. N-774 (unpublished results).

The NHase enzymes have also been found in several other microorganisms; Arthrobacter (Asano et al., 1980), Brevibacterium (Bui et al., 1982), Pseudomonas (Asano et al., 1982), Corynebacterium (Amarant et al., 1989; Tani et al., 1989), but the photoactivation phenomenon has not been observed in these enzymes. The enzyme from Brevibacterium R312 contains non-heme iron and PQQ as prosthetic groups, both of which seem to play an essential role for catalytic function (Nagasawa & Yamada, 1987; Sugiura et al., 1988). Since PQQ has so far been found only in oxidoreductases such as methylamine dehydrogenase, alcohol dehydrogenase, glucose dehydrogenase and amine oxidase, its role in combination with non-heme iron for the

[†] Author to whom all correspondence should be addressed.

[‡] Present address: Institute of Biotechnology, Korea University, Anamdong, Sungbuk ku, Seoul, Korea.

[§] Abbreviations used: NHase, nitrile hydratase; PQQ, pyrroloquinoline quinone; PEG, polyethylene glycol; V_m , volume per unit protein molecular mass.

hydration reaction is an interesting subject in enzymology.

With the purpose of acquiring some insight into the structural features of the NHase, which may explain its properties of catalytic reaction and photoactivation processes, we have initiated a study which attempts to elucidate its tertiary structure. The present work describes the crystallization and the crystallographic parameters of the photosensitive NHase from *Rhodococcus* sp. N-771 in its inactivated form.

NHase from The inactivated Rhodococcus sp. N-771 was purified using the methods described by Nagamune *et al.* (1990b), and the enzyme was further purified for crystallization by two-column chromatography. For the first column (Phenyl-Sepharose CL-4B, Pharmacia), the enzyme solution containing 15% (w/v) ammonium sulfate was loaded onto the column equilibrated with buffer (20 mm-phosphate. 10 mm-n-butyrie acid. 15 % ammonium sulfate (pH 7.5)) and eluted with a linear gradient of 15% to 0% ammonium sulfate. The eluted enzyme was loaded onto a second column (Bio-Gel HTP, Biorad) and eluted with a linear gradient of 5 mm to 50 mm-phosphate buffer (pH 6.8). Crystallization trials were carried out at 5°C and 20°C within the pH range from 5 to 8 using the hanging-drop vapor diffusion method with various conditions of NHase concentration (10 to 20 mg/ml) and of reagents (buffers, cations of chloride salt and precipitants) and their concentrations in the drop. The concentration of the precipitant in the reservoir solution was twice that in the drop. All procedures of protein purification and crystallization were performed in the dark to prevent the photoactivation of inactive NHase.

Using ammonium sulfate and PEG 8000 as the precipitants, two different crystals of the NHase gradually grew at 5°C over a period of two months. One crystallized at an ammonium sulfate concentration of 20%, in 20 to 60 mm-Hepes buffer (pH 6 to 7), 10 mm-n-butyric acid, 10 to 15 mm-zinc chloride, and has a diamond shape (size 0.7 mm \times 0.5 mm \times 0.1 mm). The other crystallized at a concentration of 5% (w/v) PEG 8000, in 20 to 60 mm-cacodylate buffer (pH 6 to 6.5), 10 mm-n-butyric acid, 5 to 15 mm-calcium chloride and has a hexagonal shape (size 0.7 mm \times 0.7 mm \times 0.2 mm). No crystals grew at 20°C, most probably due to the instability of the NHase at this temperature (Nagamune *et al.*, 1990b).

Crystallographic parameters were determined by X-ray precession photography using Ni-filtered X-rays from a Rigaku rotating anode X-ray generator (60 mA, 50 kV, 4 h exposure). The hexagonalshaped crystals belong to the hexagonal space group $P6_322$ with unit cell dimensions of $a = 110^{\circ}2$ Å and $c = 412 \cdot 1$ Å (1 Å = 0.1 nm), and diffract up to 2.8 Å on photographs. The $V_{\rm m}$ value is calculated to be 2.6 Å³/Da assuming three molecules of NHase in the asymmetric unit. The diamond-shaped crystals belong to the orthorhombic space group $P2_12_12$ with dimensions $a = 117 \cdot 4$ Å, $b = 145 \cdot 7$ Å and $c = 52 \cdot 1$ Å. The $V_{\rm m}$ value is calculated to be 2.4 Å³/Da for two molecules of NHase per asymmetric unit. These $V_{\rm m}$ values are comparable to the values of 1.68 to 3.53 Å³/Da usually found for other proteins (Matthews, 1968). The diamond-shaped crystals diffract to beyond 2.5 Å resolution and are suitable for high-resolution data collection.

References

- Amarant, T., Vered, Y. & Bohak, Z. (1989). Substrates and inhibitors of nitrile hydratase and amidase of *Corynebacterium nitrilophilus*. Biotech. Appl. Biochem. 11, 49-59.
- Asano, Y., Tani, Y. & Yamada, H. (1980). A new enzyme "nitrile hydratase" which degrades acetonitrile in combination with amidase. Agric. Biol. Chem. 44, 2251–2252.
- Asano, Y., Yasuda, T., Tani, Y. & Yamada, H. (1982). A new enzymatic method of acrylamide production. *Agric. Biol. Chem.* 46, 1183–1189.
- Bui, K., Arnaud, A. & Galzy, P. (1982). A new method to prepare amides by bioconversion of corresponding nitriles. *Enzyme Microb. Technol.* 4, 195–197.
- Ikehata, O., Nishiyama, M., Horinouchi, S. & Beppu, T. (1989). Primary structure of nitrile hydratase deduced from the nucleotide sequence of a *Rhodococcus* species and its expression in *Escherichia* coli. Eur. J. Biochem. 181, 563-570.
- Matthews, B. W. (1968). Solvent content of protein crystals. J. Mol. Biol. 33, 491-497.
- Nagamune, T., Kurata, H., Hirata, M., Honda, J., Hirata, A. & Endo, I. (1990a). Photosensitive phenomena of nitrile hydratase of *Rhodococcus* sp. N-771. *Photochem. Photobiol.* 51, 87–90.
- Nagamune, T., Kurata, H., Hirata, M., Honda, J., Koike, H., Ikeuchi, M., Inoue, Y., Hirata, A. & Endo, I. (1990b). Purification of inactivated photoresponsive nitrile hydratase. *Biochem. Biophys. Res. Commun.* 168, 437–442.
- Nagasawa, T. & Yamada, H. (1987). Nitrile hydratase is a quinoprotein. Biochem. Biophys. Res. Commun. 147. 701-709.
- Sugiura, Y., Kuwahara, J., Nagasawa, Y. & Yamada, H. (1988). Significant interaction between low-spin iron(III) site and pyrroloquinoline quinone in active center of nitrile hydratase. *Biochem. Biophys. Res. Commun.* 154, 522–528.
- Tani, Y., Kurihara, M., Nishise, H. & Yamamoto, K. (1989). Bioconversion of dinitrile to mononitrile, a tranexamic acid intermediate. by *Corynebacterium* sp. *Agric. Biol. Chem.* 53, 3143–3149.

Edited by A. Klug