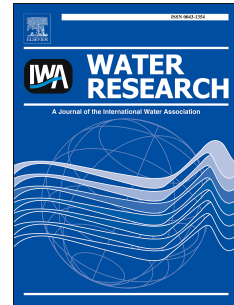


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# 1 Predicting long term removal of heavy metals from porous pavements 2 for stormwater treatment

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## 9 Abstract

10 Porous pavements are commonly used stormwater management systems. However, the understanding  
11 of their long-term capacity to retain heavy metals is limited. This study aims to investigate the long-  
12 term removal of heavy metals in three different porous pavements – Porous Asphalt (PA), Hydrapave  
13 (HP) and Permapave (PP) over accelerated laboratory experiments representing 26 years with varying  
14 hydrological conditions (drying/wetting periods and flow rates). A treatment model that simulates  
15 adsorption and desorption processes was developed for the first time to predict the long-term heavy  
16 metal removal by porous pavements. Unsurprisingly, all tested porous pavements performed better in  
17 removing metals that tend to attach to solid particles (*e.g.* Pb, Al, Fe) than more soluble ones (*e.g.* Cu,  
18 Zn, and Mn). There was a general increase of heavy metal concentrations at the outlet of the  
19 pavements over time as a result of a decrease in adsorption capacity of the systems, especially after  
20 the occurrence of clogging; the soluble heavy metals removal decreased with a reduction in flow rates  
21 which was speculated to be due to more time being available for desorption of metals and breakdown  
22 of accumulated sediments. The proposed model simulated the trend, fluctuations and peaks of heavy  
23 metal concentrations reasonably well, achieving the Nash-Sutcliffe coefficient (NSE) values of 0.53-  
24 0.68 during model calibration. The model was most promising in predicting Al and Cu release from  
25 porous pavements (50%-91% of the observed data within the 90% uncertainty bands, NSE=0.44-0.74),  
26 followed by Fe and Pb (27-77% observations within the bands, NSE=0.20-0.69). Further  
27 improvements of the model are needed for it to be applicable for Zn and Mn.

28 **Keywords:** k-C\* model; process-based model; clogging; adsorption; desorption

## 29 **1. Introduction**

30 Due to the increase in impervious areas alongside rapid urbanisation, urban stormwater runoff and  
31 pollution have increased significantly (Goonetilleke *et al.*, 2005;Zgheib *et al.*, 2012). This causes  
32 adverse impacts not only on downstream water quality (Jeng *et al.*, 2005), but also on stream health  
33 (Booth and Jackson, 1997). Meanwhile stormwater can also be an alternative resource if collected and  
34 treated properly. To manage stormwater issues in cities, a variety of techniques have been developed  
35 under the concept of Water Sensitive Urban Design (WSUD, also called Low Impact Development in  
36 USA, Sustainable Urban Drainage Systems in the UK, and Sponge City in China - Fletcher *et al.*  
37 (2015)). Porous pavements are one WSUD technology that can be easily retrofitted within dense  
38 urban areas, providing unique opportunities to infiltrate stormwater on site as source control measures  
39 without taking up space in urban landscape (Mullaney and Lucke, 2014).

40 Previous studies of the porous pavements have largely focused on their hydraulic performance (Bean  
41 *et al.*, 2007;Pezzaniti *et al.*, 2009). Indeed, the ability of porous pavement in reducing peak flow  
42 discharges and runoff volumes through filtration to the surrounding soils are the major reasons for  
43 their widespread adoption around the world (Scholz and Grabowiecki, 2007;Mullaney and Lucke,  
44 2014). Clogging (*i.e.* the decrease of its infiltration capacity) is a problem that must be considered if  
45 permeable pavements are demanded to be used as an alternative to traditional drainage systems. For  
46 example, Brattebo and Booth (2003) tested the long term infiltration capacity of four permeable  
47 pavement systems in Pacific Northwest and found they were able to infiltrate virtually all  
48 precipitations, even during the most intense stormwater (121 mm rainfall over 72 hours). Yong *et al.*  
49 (2013) studied the clogging of three permeable pavements using accelerated laboratory experiments;  
50 results show that clogging of porous pavements varied not only by their design (Porous Asphalt  
51 clogged on surface layer while Hydrapave clogged at the geotextile layer), but also subject to the  
52 operational conditions (systems exposed to drying periods have longer lifespan).

53 Porous pavements are usually regarded as being successfully in removing pollutants by adsorption,  
54 filtering and biological decomposition (Beecham *et al.*, 2012;Imran *et al.*, 2013). Heavy metals are  
55 one of the major concerns due to their acute toxicity and long-term accumulation and persistence.  
56 Pagotto *et al.* (2000) tested a porous asphalt pavement at a French highway and found 74% Pb, 62%  
57 Cd, 59% Zn and 20% Cu were removed; the authors argued that higher particulate percentage of  
58 heavy metals got more removal. 38.9% Zn, 18.2% Ni and 9.4% Pb were removed on permeable  
59 pavement made of 20-mm grave sub-base (280 mm high) over several rain events in a car park of  
60 south Australia (Beecham *et al.*, 2012). Myers *et al.* (2011) assessed the impact of residence time on  
61 heavy metal retention on permeable pavement with quartzite and dolomite as base material during a  
62 large simulated event; they discovered that Zn, Cu and Pb removal was between 94 and 99% after 144  
63 h of retention in the base layer, but the removal was lower (~61% Zn, 35% Pb and 30% Cu) during  
64 the initial stages where the residence time was only 1 hour. Dierkes *et al.* (2002) used accelerated  
65 experiments to test four different types of pavers at a rainfall intensity of 144 mm/hr as worst case  
66 scenario simulating 5 years of rain in Germany, results show that 89-98% Pb, 74-98% Cd, 89-96%  
67 and 72-97% Zn were removed, respectively; same study also suggested that basalt and gravel as  
68 subbase materials are better in removing heavy metals than limestone and sandstone materials. A  
69 recent study by Sountharajah *et al.* (2017) found that using zeolite or basalt as bed material in  
70 porous pavements removed 41-72% Cd, 67-74% Cu, 38-43% Ni, 61-72% Pb and 63-73% Zn  
71 respectively during accelerated 80h period experiment that simulated 10 years of Sydney rainfall  
72 using uniform distribution of rainfall.

73 The methodologies used in the above studies were mostly simple short-term field or accelerated  
74 laboratory studies on relatively new systems, which failed to consider the impact of highly variable  
75 operational conditions (e.g. dry/wetting periods between events and varying flow rates) over life span  
76 of these systems. Brattebo and Booth (2003) conducted a rare long-term experiment (over six-year  
77 operation) on a heavily used porous pavement in a parking area, and found that both positive and  
78 negative changes of released heavy metal concentrations: Zn outflow concentration increased from 5  
79  $\mu\text{g/L}$  to 10  $\mu\text{g/L}$ , while that of Cu decreased from 10  $\mu\text{g/L}$  to  $< 3 \mu\text{g/L}$  during the six-year study

80 period. It can be concluded that, although life span of porous pavements can go well over 25 years,  
81 the knowledge on how these systems perform in removing heavy metals over long time periods is still  
82 limited. Additionally, there is no specific study that investigates heavy metal removal processes  
83 within porous pavements which may help to understand the long-term removal performance.

84 There are models available to simulate the hydraulic behaviour of porous pavements; *e.g.* in the  
85 commercially available software SWMM by USEPA (Rossman, 2017), a porous pavement system is  
86 modelled as an infiltration system that combines three vertical layers (*i.e.* the surface, pavement and the  
87 storage layers). The method has also been tested by others to understand the hydraulic performance of  
88 permeable pavement systems (Zhang and Guo, 2015). To account for the clogging process that is  
89 often observed in porous pavements, Yong *et al.* (2013) proposed a simple four-parameter black-box  
90 regression model that for the first time predicts physical clogging as a function of cumulative volume  
91 and climatic conditions.

92 Unfortunately, there is a lack of algorithms that can simulate the pollution treatment processes within  
93 porous pavement systems. The first order kinetic decay model (also called k-C\* model), serves the  
94 mostly widely used method that has also been adopted in software packages such as SWMM  
95 (Rossman, 2017) and MUSIC by eWater (eWater, 2014). However, the inadequacies of k-C\* model  
96 are often mentioned due to its simplicity (*e.g.* assumption of constant k and C\* value) (Kadlec and  
97 Knight, 1996; Newton, 2005). Newton (2005) successfully used a one-parameter first decay model  
98 adapted from filtration theory for wastewater treatment to predict particle removal efficiency from  
99 pavement with satisfactory, *e.g.* NSE=0.36-0.98 for low flow rates and from negative to 0.39 for high  
100 flow rates. Both empirical models and conceptual model (adapted from a sediment removal model for  
101 a sand filter) were developed by He *et al.* (2015) to predict suspended solids and phosphorus removal by  
102 a porous concrete pavement; the prediction errors were within 5.29% for two validation events. These  
103 models are however mainly for event-based predictions and do not account for specific treatment  
104 processes (*e.g.* adsorption & desorption); they are also developed for mainly sediments and nutrients,  
105 not suitable for heavy metals that undergo via different removal mechanisms. Hence development of a  
106 process-based water quality model that not only involves key treatment processes but also can

107 simulate long-term treatment performance of heavy metal by porous pavements is required to assist in  
108 better designs of these systems.

109 This paper aims to fill in these knowledge gaps, firstly by understanding heavy metal removal  
110 performance of three different porous pavements (porous asphalt, hydrapave and permapave) over a  
111 long term under different conditions, and then developing for the first time a model that not only  
112 predicts long-term heavy metal removal but also explains the removal processes. The specific  
113 objectives of this study are to:

- 114 • test the treatment performance of the three porous pavements for different heavy metals (Al, Cd,  
115 Cr, Cu, Fe, Mn, Ni, Pb and Zn) using accelerated laboratory experiments spanning over 1 year  
116 representing 26 years of operation under varying operational conditions;
- 117 • understand the impact of clogging, pavement type and flow rate on treatment performance; and
- 118 • develop, test and validate a treatment model accounting for main removal processes (e.g.  
119 adsorption and desorption) for prediction of long term removal of heavy metals;

120 We hypothesis that heavy metals will accumulate in the system and also get released over time from  
121 the systems, and the metal characteristics, pavement design, and hydrological conditions are the key  
122 influential factors. The proposed model accounting for heavy metal adsorption and desorption will be  
123 able to provide reasonable predictions for majority of the tested heavy metals but not good for some  
124 that have other removal processes.

## 125 2. Methods

### 126 2.1 Experimental set-up

127 Three porous pavement systems that are commercially available were used in this study:

- 128 • **monolithic porous asphalt (PA)** – a standard bituminous asphalt surface (40mm), underlaid  
129 by a layer of crushed aggregate (40 mm), and a highly permeable layer of open graded clean  
130 washed aggregate with >40% void space as reservoir bed (570 mm);

- 131 • **modular Hydrapave (HP)** – a thick paver made of Boral clay and concrete (80 mm), which  
132 is laid on  $\Phi 5$  mm clean stone (50mm), a geotextile layer, and another two sublayers of  $\Phi 5$ -20  
133 mm stone (100 mm) and  $\Phi 10$ -63 mm stone (250 mm);
- 134 • **Permapave (PP)** – a thick paver of  $\Phi 10$ -12 mm crashed gravel (50 mm), underlaid by a sub-  
135 base layer of  $\Phi 5$ -20 washed gravel (350 mm).

136 We used the same experimental rig (Figure 1) that has been employed in the parallel studies of the  
137 clogging and nutrient removal by the porous pavements, as reported in Yong *et al.* (2013). The rig had  
138 a 550 L tank with constant mixing, from which the inflow is evenly distributed via a distribution  
139 system (peristaltic pump + rotating sprinkler) into three separate vertical compartments representing  
140 three different pavements (each has a size of  $0.9 \times 0.45 \times 1.95$  m); three separate tipping bucket rain  
141 gauges (0.2 mm/tip resolution) were installed at the end of the system to monitor the outflow rates.  
142 Results from the clogging study (Yong *et al.*, 2013) have shown that PA and HP exhibited initial  
143 clogging (*i.e.* the ponding above the pavement surfaces overflows) after 11 years and 12 year  
144 respectively of accelerated operations under various drying and wetting conditions, while PP had no  
145 sign of clogging after 26 years. All the three systems had good performance in removing sediments,  
146 but had varying performance for nutrients removal depending on the flow rates (Yong *et al.*, 2011).

147  
148 **Figure 1** The experimental set-up for testing Porous Asphalt, Hydrapave and Permapave (adapted  
149 from Yong *et al.* (2013))

## 150 2.2 Experimental procedure

### 151 2.2.1 Inflow synthetic stormwater

152 Semi-synthetic stormwater was prepared in the 550 L tank according to the methods described  
153 previous in stormwater studies (Blecken *et al.*, 2009), with standard Australia stormwater quality  
154 (Duncan, 1999). The target concentrations of sediments, nutrients and heavy metals in semi-synthetic  
155 stormwater are presented in Table 1, together with the primary source of the pollutants.

156 **Table 1 Semi-synthetic stormwater water quality**

Pollutant	Target concentration	Primary source of pollutant added
Total suspended solids (TSS)	150 mg/L	Stormwater wetland sediment
Total Nitrogen (TN)	2.1 mg/L	KNO <sub>3</sub> , NH <sub>4</sub> CL, C <sub>6</sub> H <sub>5</sub> O <sub>2</sub> N, wetland Sediment
Total Phosphorus (TP)	0.35 mg/L	KH <sub>2</sub> PO <sub>4</sub>
Aluminium (Al)	4.0 mg/L	standard solution
Cadmium (Cd)	0.0045 mg/L	standard solution
Chromium (Cr)	0.025 mg/L	Cr(NO <sub>3</sub> ) <sub>3</sub>
Copper (Cu)	0.05 mg/L	CuSO <sub>4</sub>
Iron (Fe)	3.0 mg/L	standard solution
Manganese (Mn)	0.25 mg/L	Mn(NO <sub>3</sub> ) <sub>2</sub>
Nickel (Ni)	0.03 mg/L	Ni(NO <sub>3</sub> ) <sub>2</sub>
Lead (Pb)	0.14 mg/L	Pb(NO <sub>3</sub> ) <sub>2</sub>
Zinc (Zn)	0.25 mg/L	ZnCl <sub>2</sub>

157 **2.2.2 Dosing of the system under varying wetting/drying regimes**

158 Over a course of one year, 26 years of operation in a typical sub-tropical Brisbane climate (average  
 159 annual rainfall – 1200 mm) was simulated, under various wetting/drying conditions. Four inflow rates  
 160 were simulated (Table 2), with flow A, B, C and D representing the average rainfall intensity of the 0-  
 161 39, 40-59, 60-79 and 80-100 percentile groups, respectively; in addition, a 1 in 5-year design storm  
 162 over 5 minutes was also chosen to simulate the typical design storm for small catchments where  
 163 porous pavements are likely to be installed. These flows were estimated from the Brisbane runoff-  
 164 frequency curve, which was generated using MUSIC model (eWater, 2014) and six-minute rainfall  
 165 data collected between 1988 and 1997 in Brisbane.

166 **Table 2 System inflow rates used in the experiment**

Flow	Frequency (percentile range)	Flow rate (L/h/ha)	Velocity (mm/h)	Number of times flow rate was simulated	Duration of inflow each time flow was simulated (h)
A	0-39	0.6	0.2	26	96
B	40-59	2.9	1.0	26	48
C	60-79	7.1	2.6	26	48
D	80-100	60.9	21.9	26	48
1 in 5-yr storm	-	530	191	6 <sup>a</sup>	5

167 <sup>a</sup> Occurred in Year 5.9, 8.1, 11.8, 15.6 19.5 and 23.5.

168 Generally, each simulated year consisted of four flow types: A, B, C and D, which were applied for  
 169 96, 48, 48 and 48 h respectively (48 h represents approximately 52 simulated days, note each flow  
 170 was not applied continuously but with many dry periods – see next paragraph for details); the total  
 171 amount of applied annual inflow was 1243 mm (close to Brisbane annual rainfall). The order of the



172 flow types was applied randomly, *e.g.* in year 1, the sequence of D, C, B, A may be applied, while in  
173 Year 2 it may become the sequence of C, A, B, D. The 1 in 5-year stormwater events were simulated  
174 in Year 5.9 (Storm 1), 8.1 (Storm 2), 11.8 (Storm 3), 15.6 (Storm 4), 19.5 (Storm 5) and 23.5 (Storm  
175 6).

176 To account for the drying, the inflow was not applied continuously, but with dry periods in-between  
177 each event. According to the methods described in our previous work (Yong *et al.*, 2013), it was  
178 determined that an average of 21 dry weather periods occurred during any given year in Brisbane. As  
179 such, in each simulated year, 21 dry periods were mimicked by applying fan heaters at 25 °C for 3 h  
180 (which removed 80% of the moisture content in the pavements that is equivalent to 4 days of natural  
181 dry – this was determined through a preliminary experiment).

### 182 2.2.3 Sampling and analysis

183 For each flow rate, three time-weighted samples were collected at both inflow and outflow point over  
184 the entire duration of the event to form two composite samples (*i.e.* one inflow and one outflow). The  
185 collection of samples was accompanied by pH measurement to enable early predictions to be made  
186 about the behaviour of heavy metals in the systems. Once collected, the samples were acidified, stored  
187 in fridge and then delivered to a NATA accredited laboratory for analysis of nine heavy metals in  
188 accordance with the standard methods described in APHA-AWWA-WPCF (2005): Aluminium (Al),  
189 Cadmium (Cd), Chromium (Cr), Copper (Cu), Iron (Fe), Manganese (Mn), Nickel (Ni), Lead (Pb) and  
190 Zinc (Zn); the LOR (limit of report) was 0.01 mg/L for Al and Fe and 0.001 mg/L for the rest.

## 191 2.3 Long term treatment model development

### 192 2.3.1 Proposed model algorithms

193 In this study the simple first order decay model ( $k - C^*$  model, Kadlec and Knight (1996)) is adapted  
194 with revisions to include adsorption and desorption processes for simulation of the long-term of heavy  
195 metals from porous pavements. The basic equation of the  $k - C^*$  model is:

$$\frac{C_{out} - C^*}{C_{in} - C^*} = e^{-\frac{k}{q}} \quad 1$$

196 where  $C_{in}$  - inflow concentration, mg/L;  $C_{out}$  – outflow concentration, mg/L;  $C^*$  - the background  
 197 concentration, mg/L;  $k$  – the event decay parameter, day/L; and  $q$  is the hydraulic loading (in this case  
 198 flow rate, L/day).

199 Equation 1 can be rearranged and written in time-step basis for estimating  $C_{out}$ , as:

$$C_{out}(t) = C^* + [C_{in}(t) - C^*] e^{-\frac{k}{q(t)}} \quad 2$$

200 The background concentration  $C^*$  is often used as a constant parameter (*e.g.* in MUSIC, pre-calibrated  
 201  $C^*$  values are used for treatment performance modelling for all the treatment measures (eWater,  
 202 2014)). However, we hypothesised that  $C^*$  is not constant, and may (1) decrease due to adsorption  
 203 process – depending on inflow (as bench marking concentration) and adsorption rate ( $k_{ad}$ ), and (2)  
 204 increase due to desorption process – depending on the total amount of pollutant accumulated in the  
 205 previous time step ( $M(t-1)$ , g) and desorption rate ( $k_{des}$ , 1/L). So we proposed that:

$$C^*(t) = [C_{in}(t) - k_{ad} C_{in}(t)] + k_{des} M(t-1) \quad 3$$

206 Hence, the outflow concentration ( $C_{out}$ ) can be estimated using Equation 4 and 5:

$$C_{out}(t) = C_{in}(t) - k_{ad} C_{in}(t) + k_{des} M(t-1) + [k_{ad} C_{in}(t) - k_{des} M(t-1)] e^{-\frac{k}{q(t)}} \quad 4$$

$$M(t) = M(t-1) + [q_{in}(t)C_{in}(t) - q_{out}(t)C_{out}(t)]dt \quad 5$$

207 The model has three parameters: the event decay rate ( $k$ ), the adsorption rate ( $k_{ad}$ ) and desorption rate  
 208 ( $k_{des}$ ). The initial condition is  $M|_{t=0} = 0$ .

209 2.3.2 *Data preparation, model calibration and validation*

210 The model was tested only for Hydrapave (HP) and Porous Asphalt (PA); Permapave was excluded  
211 for model testing as its outflow rates were not measured properly due to the failure of the rain gauge.

212 During the experiment, inflow rates were controlled (Table 2) while the outflow rates were measured  
213 using tipping-bucket rain gauge (0.2 mm/tip), the flow rates were then prepared in hourly time-steps  
214 (equivalent to 1.08 simulated day, *i.e.* approximately daily time-step). However, water quality  
215 samples were not collected on hourly time-steps, but as 48 hours (52 simulated days) composite  
216 samples (see Section 2.2.3). It was therefore assumed that the concentrations within each 48 hours  
217 period did not change; *i.e.* concentrations at any hour within the period were assumed to be the same  
218 as the measured composite concentration for the 48 hour period. In this way, inflow and outflow rate,  
219 as well as heavy metal concentrations were prepared on an hourly time-step (*i.e.* simulated daily time-  
220 step) for the proposed model testing.

221 The model was run in a simulated daily time-step for the first half of the experiment (*i.e.* simulated  
222 Year 1-13 for HP and Year 1-10 for PA) for model calibration. At the middle of the time-step when a  
223 composite sample was collected, the simulated concentration was extracted; *i.e.* if the composite  
224 sample was taken from Hour 1- Hour 48 (excluding the drying period), the simulated concentration is  
225 extracted at Hour 24. All the extracted concentrations from simulation were compared to the  
226 concentrations at that time-step (as observed) for model testing using the Nash-Sutcliffe coefficient –  
227 NSE (Nash and Sutcliffe, 1970). 10,000 model runs were conducted for parameter calibration, with  
228 parameters values randomly sampled from uniform distributions (the ranges were informed by  
229 preliminary model runs practices – refer to Table S1 of Supplementary Material for the detail  
230 information); the use of uniform distributions was recommended by previous studies by Freni and  
231 Mannina (2010) when there is lack of parameter information.

232 Validation of the proposed model was performed using the second half of the experiment (which is  
233 independent of the data for model calibration). Top 1% of the parameter sets (*i.e.* 100) from  
234 calibration were chosen to generate the parameter distributions, which were then used to estimate the

235 model prediction uncertainty (90 % probability bands) using GLUE method (Beven and Binley, 1992).  
236 It should be acknowledged that selection of 100 behavioural runs was quite arbitrary; it however still  
237 satisfied the minimum runs required by GLUE, and selecting the top 1% simulations resulted in much  
238 higher acceptability thresholds (*e.g.* in this paper  $NSE > 0.45$  for Al, Cu, Fe, Pb and Zn) comparing to  
239 traditional urban drainage models (*i.e.* 0.0); Freni *et al.* (2008) also suggested that higher thresholds  
240 not only allow for obtaining more relevant information of parameters responsibility in modelling  
241 uncertainty propagation but also allow for a stricter verification of the model. The thresholds for Mn  
242 were however only NSE of 0.10 for HA and  $<0.0$  for PA, the uncertainty analysis was anyway  
243 proceeded using the top 1% parameter sets for Mn.

### 244 3. Results and discussion

#### 245 3.1 Treatment performance

##### 246 3.1.1 Overall performance

247 The metal treatment performance of the three porous pavements over 26 simulated years are  
248 summarized in Table 3. All three pavements had cumulative heavy metals removal rates of over 50%;  
249 they were the most effective for Pb ( $84\pm 14\%$ ), Al ( $79\pm 13\%$ ), Fe ( $77\pm 13\%$ ), and less for Cu ( $68\pm 19\%$ )  
250 and Zn ( $66\pm 20\%$ ). Lower but highly variable removal was found for Mn ( $35\pm 35\%$ ), while for Ni, net  
251 productions were observed in most of the cases which could be entirely due to uncertainty in  
252 measurements of very low inflow concentration (often of  $<0.005$  mg/L). Cd and Cr also had very low  
253 inflow concentrations ( $<0.01$  mg/L) and were mostly non-detected in outflow samples; as such, Cd,  
254 Cr and Ni were then excluded from future discussions. These findings have good agreement with  
255 previous studies (Pagotto *et al.*, 2000; Sounthararajah *et al.*, 2017). The different performance between  
256 heavy metals can be explained by their affinity to particulates; *e.g.* Pb, Al and Fe are easily attached to  
257 sediments in stormwater (Makepeace *et al.*, 1995) and hence can be readily retained when filtering  
258 through porous pavements; these retained metals can form stable complexes via surface complexation  
259 reactions (Bradl, 2004). Cu and Zn are largely presented in dissolved form (Makepeace *et al.*, 1995)  
260 and their retention by porous pavements usually undergo via rather weak processes such as ion

261 exchange. Mn also has good attachment with sediments/organic matter in stormwater, and in general  
262 the removal process of Mn can be very complex in the form of Mn oxides (Bradl, 2004); the low and  
263 variable removal observed in this study was probably due to the very low inflow concentrations (*i.e.*  
264  $0.0228 \pm 0.006$  mg/L for Mn and  $0.0032 \pm 0.001$  mg/L for Ni).

265 When considering reusing treated stormwater or discharging to protect eco-system health, the degrees  
266 of heavy removal by porous pavements were insufficient since the effluent heavy metal mostly exceed  
267 Australia Drinking Water Guidelines (NHMRC-NRMMC, 2011) and trigger values for aquatic health  
268 protect and irrigation, except for Mn (Table 3). Hence it is suggested further treatment through Water  
269 Sensitive Urban Design systems (*e.g.* stormwater biofilters) (Payne *et al.*, 2015) or some advanced  
270 technologies such as using nano-fibrous material filtrations shall be used (Sounthararajah *et al.*, 2017).

271 **Table 3** Performance of three different porous pavements over 26 simulated years: average inflow and outflow concentrations before clogging (BS) and after  
 272 clogging (AC)

Flow	Al			Cd			Cr			Cu			Fe			Mn			Ni			Pb			Zn				
	In	Out	%	In	Out	%	In	Out	%	In	Out	%	In	Out	%	In	Out	%	In	Out	%	In	Out	%	In	Out	%		
Porous pavement (PP)	A	BC	3.45	0.324	<b>91</b>	0.006	N.D.	-	0.005	N.D.	-	0.270	0.063	<b>77</b>	2.98	0.383	<b>87</b>	0.024	0.018	<b>25</b>	0.004	0.012	<b>-200</b>	0.110	0.002	<b>98</b>	0.391	0.113	<b>71</b>
		AC	1.19	0.453	<b>62</b>	0.003	N.D.	-	N.D.	N.D.	-	0.079	0.086	<b>-9</b>	1.02	0.505	<b>50</b>	0.008	0.017	<b>-113</b>	0.001	0.006	<b>-500</b>	0.023	0.003	<b>87</b>	0.238	0.217	<b>9</b>
	B	BC	4.34	0.492	<b>89</b>	0.006	N.D.	-	0.006	N.D.	-	0.29	0.068	<b>77</b>	3.75	0.505	<b>87</b>	0.025	0.018	<b>28</b>	0.005	0.010	<b>-100</b>	0.132	0.002	<b>98</b>	0.397	0.092	<b>77</b>
		AC	3.02	0.581	<b>81</b>	0.005	N.D.	-	0.004	N.D.	-	0.180	0.080	<b>56</b>	2.55	0.564	<b>78</b>	0.016	0.018	<b>-13</b>	0.002	0.005	<b>-150</b>	0.072	0.006	<b>92</b>	0.309	0.207	<b>33</b>
	C	BC	5.83	0.692	<b>88</b>	0.007	N.D.	-	0.009	N.D.	-	0.355	0.069	<b>81</b>	4.90	0.637	<b>87</b>	0.033	0.018	<b>45</b>	0.004	0.009	<b>-125</b>	0.148	0.002	<b>99</b>	0.466	0.101	<b>78</b>
		AC	6.38	0.574	<b>91</b>	0.006	N.D.	-	0.009	N.D.	-	0.309	0.087	<b>72</b>	5.44	0.582	<b>89</b>	0.029	0.021	<b>28</b>	0.003	0.006	<b>-100</b>	0.155	0.004	<b>97</b>	0.402	0.186	<b>54</b>
	D	BC	5.40	1.24	<b>77</b>	0.007	N.D.	-	0.008	N.D.	-	0.303	0.077	<b>75</b>	4.44	1.05	<b>76</b>	0.029	0.019	<b>34</b>	0.005	0.006	<b>-20</b>	0.134	0.013	<b>90</b>	0.430	0.163	<b>62</b>
		AC	4.92	1.21	<b>75</b>	0.006	N.D.	-	0.007	N.D.	-	0.267	0.089	<b>67</b>	4.07	1.02	<b>75</b>	0.022	0.016	<b>27</b>	0.002	0.005	<b>-150</b>	0.129	0.013	<b>90</b>	0.367	0.191	<b>48</b>
	5-yr	BC	5.73	2.78	<b>51</b>	0.007	N.D.	-	0.008	0.004	<b>50</b>	0.333	0.160	<b>52</b>	5.07	2.48	<b>51</b>	0.028	0.025	<b>11</b>	0.005	0.004	<b>20</b>	0.135	0.058	<b>57</b>	0.436	0.237	<b>46</b>
		AC	4.75	2.18	<b>54</b>	0.006	0.003	<b>50</b>	0.007	0.003	<b>57</b>	0.250	0.132	<b>47</b>	4.04	1.81	<b>55</b>	0.023	0.017	<b>26</b>	0.004	0.004	<b>0</b>	0.148	0.066	<b>55</b>	0.373	0.211	<b>43</b>
Total	BC	4.80	0.778	<b>84</b>	0.007	N.D.	-	0.007	N.D.	-	0.305	0.073	<b>76</b>	4.06	0.723	<b>82</b>	0.028	0.019	<b>32</b>	0.004	0.009	<b>-125</b>	0.131	0.007	<b>95</b>	0.422	0.122	<b>71</b>	
	AC	4.01	0.937	<b>77</b>	0.005	N.D.	-	0.005	N.D.	-	0.215	0.093	<b>57</b>	3.39	0.847	<b>75</b>	0.019	0.018	<b>5</b>	0.002	0.005	<b>-150</b>	0.103	0.016	<b>84</b>	0.336	0.202	<b>40</b>	
Hydrapave (HP)	A	BC	2.89	0.097	<b>97</b>	0.005	N.D.	-	0.004	N.D.	-	0.218	0.046	<b>79</b>	2.35	0.206	<b>91</b>	0.020	0.017	<b>15</b>	0.002	0.011	<b>-450</b>	0.089	0.002	<b>98</b>	0.347	0.068	<b>80</b>
		AC	1.30	0.408	<b>69</b>	0.004	N.D.	-	0.003	N.D.	-	0.096	0.060	<b>38</b>	1.11	0.409	<b>63</b>	0.009	0.008	<b>11</b>	0.002	0.007	<b>-250</b>	0.041	0.007	<b>83</b>	0.249	0.118	<b>53</b>
	B	BC	4.53	0.124	<b>97</b>	0.006	N.D.	-	0.006	N.D.	-	0.288	0.039	<b>86</b>	3.85	0.201	<b>95</b>	0.026	0.017	<b>35</b>	0.003	0.011	<b>-267</b>	0.127	0.002	<b>98</b>	0.405	0.061	<b>85</b>
		AC	2.78	0.345	<b>88</b>	0.005	N.D.	-	0.003	N.D.	-	0.187	0.056	<b>70</b>	2.35	0.354	<b>85</b>	0.015	0.009	<b>40</b>	0.002	0.008	<b>-300</b>	0.080	0.005	<b>94</b>	0.308	0.136	<b>56</b>
	C	BC	4.87	0.205	<b>96</b>	0.006	N.D.	-	0.007	N.D.	-	0.302	0.037	<b>88</b>	4.10	0.284	<b>93</b>	0.027	0.017	<b>37</b>	0.003	0.008	<b>-167</b>	0.133	0.002	<b>98</b>	0.406	0.052	<b>87</b>
		AC	4.73	0.547	<b>88</b>	0.006	N.D.	-	0.006	N.D.	-	0.244	0.054	<b>78</b>	3.99	0.498	<b>88</b>	0.023	0.008	<b>65</b>	0.004	0.006	<b>-50</b>	0.124	0.012	<b>90</b>	0.380	0.116	<b>69</b>
	D	BC	5.18	1.053	<b>80</b>	0.006	N.D.	-	0.007	N.D.	-	0.292	0.062	<b>79</b>	4.36	0.892	<b>80</b>	0.027	0.013	<b>52</b>	0.003	0.004	<b>-33</b>	0.137	0.011	<b>92</b>	0.411	0.106	<b>74</b>
		AC	4.07	0.997	<b>76</b>	0.006	N.D.	-	0.006	N.D.	-	0.246	0.062	<b>75</b>	3.42	0.813	<b>76</b>	0.020	0.008	<b>60</b>	0.004	0.004	<b>0</b>	0.127	0.022	<b>83</b>	0.363	0.128	<b>65</b>
	5-yr	BC	5.63	2.17	<b>62</b>	0.007	N.D.	-	0.009	0.006	<b>33</b>	0.333	0.122	<b>63</b>	5.04	1.98	<b>51</b>	0.028	0.015	<b>46</b>	0.005	0.007	<b>-40</b>	0.136	0.039	<b>71</b>	0.429	0.160	<b>63</b>
		AC	4.65	2.11	<b>55</b>	0.006	N.D.	-	0.008	N.D.	-	0.270	0.120	<b>56</b>	3.93	1.69	<b>57</b>	0.023	0.012	<b>48</b>	0.003	0.003	<b>0</b>	0.159	0.072	<b>55</b>	0.379	0.157	<b>59</b>
Total	BC	4.43	0.448	<b>90</b>	0.006	N.D.	-	0.006	0.002	<b>67</b>	0.277	0.049	<b>82</b>	3.73	0.464	<b>88</b>	0.025	0.016	<b>36</b>	0.003	0.008	<b>-167</b>	0.122	0.005	<b>96</b>	0.394	0.076	<b>81</b>	
	AC	3.40	0.766	<b>77</b>	0.005	N.D.	-	0.005	0.002	<b>60</b>	0.203	0.066	<b>67</b>	2.87	0.665	<b>77</b>	0.018	0.009	<b>50</b>	0.003	0.006	<b>-100</b>	0.101	0.019	<b>81</b>	0.332	0.129	<b>61</b>	
Permapave (PP)	A		2.34	0.233	<b>90</b>	0.005	N.D.	-	0.003	N.D.	-	0.172	0.031	<b>82</b>	2.10	0.291	<b>86</b>	0.016	0.004	<b>76</b>	0.002	0.002	<b>39</b>	0.068	0.002	<b>97</b>	0.316	0.023	<b>93</b>
	B		3.90	0.634	<b>84</b>	0.006	N.D.	-	0.006	N.D.	-	0.256	0.052	<b>80</b>	3.40	0.652	<b>81</b>	0.022	0.009	<b>58</b>	0.003	0.001	<b>59</b>	0.115	0.007	<b>94</b>	0.376	0.031	<b>92</b>
	C		5.11	0.589	<b>88</b>	0.006	N.D.	-	0.008	N.D.	-	0.282	0.046	<b>84</b>	4.46	0.589	<b>87</b>	0.028	0.006	<b>78</b>	0.004	0.001	<b>73</b>	0.128	0.004	<b>97</b>	0.388	0.022	<b>94</b>
	D		4.93	1.17	<b>76</b>	0.006	N.D.	-	0.007	N.D.	-	0.279	0.064	<b>77</b>	4.12	0.996	<b>76</b>	0.025	0.007	<b>71</b>	0.003	0.001	<b>69</b>	0.137	0.015	<b>89</b>	0.397	0.046	<b>88</b>
	5-yr		4.94	2.28	<b>54</b>	0.006	N.D.	-	0.007	N.D.	-	0.289	0.121	<b>58</b>	4.28	1.89	<b>56</b>	0.024	0.010	<b>57</b>	0.003	0.002	<b>53</b>	0.144	0.058	<b>60</b>	0.388	0.120	<b>69</b>
Total		4.14	0.78	<b>81</b>	0.006	N.D.	-	0.006	N.D.	-	0.250	0.054	<b>78</b>	3.58	0.73	<b>80</b>	0.023	0.007	<b>70</b>	0.003	0.001	<b>61</b>	0.115	0.011	<b>91</b>	0.371	0.037	<b>90</b>	
ANZECC <sup>2</sup>	ADWG <sup>1</sup>		<i>0.2</i>		0.002			0.05		1			<i>0.3</i>			0.1			0.02			<i>0.01</i>				3			
	Fresh		<i>0.055</i>		0.0002			0.001		<i>0.0014</i>			-			1.9			<i>0.0011</i>			<i>0.0034</i>				<i>0.008</i>			
	Marine		-		0.0055			0.0044		<i>0.0013</i>			-			0.07			<i>0.0044</i>			<i>0.0044</i>				<i>0.015</i>			
	LTV		5		0.01			0.1		0.2			<i>0.2</i>			0.2			0.2			2				2			
	STV		20		0.05			1		5			10			10			2			5				5			

273 <sup>1</sup>ADWG - Australian Drinking Water Guideline Values (NHMRC-NRMMC, 2011); <sup>2</sup>ANZECC - Australian and New Zealand Guidelines for Fresh and Marine Water Quality (ANZECEPH&ARMCANZ, 2000):

274 Fresh and Marine means trigger values for 95% protection of species in Fresh and Marine waters, respectively; LTV and STV are long-term trigger value and short-term trigger value for heavy metals in irrigation

275 waters. Values in *italic 'Red'* means outflow concentrations of the heavy metal is above the trigger value.

## 276 3.1.2 Performance change over time and impact of clogging

277 For majority of the heavy metals, a general increasing trend of effluent concentrations with  
278 fluctuations was observed over the whole experiment (Figure 2); the fluctuations were due to the  
279 variable input flows and concentrations (that mimic reality), with six large ‘storm events’ contributed  
280 to the highest outflow peaks. With the accumulation of metals in the systems, adsorption sites became  
281 limited and desorption turned to be more prominent. Nevertheless, the outflow concentrations were  
282 still below the inflow concentrations until the end of the experiment (*e.g.* after 20-26 years of  
283 operation), indicating that these systems still have capacity for metal removal. Mn, exhibited  
284 surprisingly decreasing concentration over time (Figure 2), *e.g.* the effluent Mn concentrations were  
285 2-3 times lower in the end compared with the start. Bradl (2004) suggested that Mn oxides are good  
286 sinks for Cu and Fe oxides, and can also form Pb-Mn formation, hence exhibit co-precipitation which  
287 enhances Mn removal over time.

288 Another important finding is that in the early stage of the system (1-2 years), the systems had poorer  
289 performance and exhibited larger variabilities especially for the first few sampling events (especially  
290 for Mn which even had net productions; Figure 2); for example, outflow Cu concentrations in the first  
291 two years were  $0.076 \pm 0.048$  mg/L, which dropped to  $0.045 \pm 0.023$  mg/L over the following two  
292 years. Therefore, porous pavement systems need time to mature for stable and improved performance.  
293 Clogging, which occurred in Year 11 in PA and Year 12 in HP had obvious impact on the system  
294 performance (Figure 2; Table 3). It is estimated that after clogging average outflow concentrations  
295 were 1.2 – 2.4 times higher than that before clogging for PA systems (and 1.3 - 3.6 times for HP  
296 system). As time progressed, clogging resulted in an increase in the detention time, allowing more  
297 time for the desorption process, which became more prominent (due to the accumulated heavy metals  
298 over time) than the adsorption. This is different from previous study by Myers *et al.* (2011) who found  
299 that longer residence time led to better removal of Zn, Cu and Pb. The study however only  
300 investigated one single large event (*i.e.* short-term) on a fresh permeable pavement that had no  
301 clogging issue; this reaffirms the importance of this study which looked into the long term  
302 performance of the pavement systems in removing heavy metals.

303 *Figure 2 Change of outflow heavy metal concentrations over the course of 26 simulated years;*  
304 *“average inflow concentration  $\pm$  standard deviation” indicated in brackets of the legends.*

### 305 3.1.3 *Impact of pavement type*

306 The difference in average removal between the three types of porous pavements were very small (up  
307 to 5%; Figure 3A), with exception of the soluble metals – Cu, Zn and Mn. As discussed, system  
308 clogging had adverse impact on system performance, hence the PP (which was not clogged)  
309 performed best for removing these heavy metals; *e.g.* specifically the average outflow heavy metal  
310 concentrations from PP were only 71% and 56% of those from HP and PA, respectively (Table 3). In  
311 addition, the difference in sub-base materials may also contribute to the different observations.  
312 Dierkes *et al.* (2002) observed that paving stones of porous concrete and green apertures (similar to  
313 PP and PA in this study) had better heavy metal retention capacities than pavers with joints (similar to  
314 HP in this study). Although gravel and basalt as base material are usually more effective in removing  
315 pollutants (Dierkes *et al.*, 2002), the basalt used in PA system of this study used are much coarser than  
316 the stones in HP and gravel in PP, thus resulting the lowest removal for Cu, Zn and Mn.

### 317 3.1.4 *Impact of flow rate*

318 Figure 3B indicates the impact of flow rates on removal. As expected, the simulated flow representing  
319 1 in 5-yr storm led to the poorest performance, *e.g.* the average removal of heavy metals (except for  
320 Mn) were usually within the range of 40-55%, while it was >70% under the other flow rates. The  
321 differences of average removal rates between other flow rates (Flow A, B, C and D) were small  
322 (<10%) for particulate metals (*i.e.* Pb, Al and Fe), but higher outflow concentrations usually  
323 corresponded to larger flow rates (*e.g.* Flow D had ~ 4 times higher average outflow concentrations  
324 than Flow A; Table 3), which has good agreement with previous study on the same systems for  
325 removal of TP (however opposite trend was found for TN) (Yong *et al.*, 2011). As for the soluble  
326 heavy metals (*i.e.* Cu, Zn, and Mn), relatively higher removals were observed at Flow C; surprisingly,  
327 it was found that the lowest rate (Flow A) had the largest variability in heavy metal removal compared  
328 with other flow rates (Figure 3B), which presumably due to the big reduction in removal after



329 clogging (Table 3); in cases that clogging occurred, lower outflow rates were observed and it  
 330 benefited to the desorption process of these weakly attached metals, or breakdown of trapped  
 331 sediments in the clogging layer facilitated by longer detention (during the event) as previous reported  
 332 to impact TN outflow concentration from the systems (Yong *et al.*, 2011); both processes may  
 333 introduce more uncertainties and result in the higher variability.

334 *Figure 3 Influence of (A) pavement types and (B) flow rates on heavy metal removal*

### 335 3.2 Model testing results

336 The performance of the proposed model and values of calibrated parameters are summarized in Table  
 337 4, with the observed and simulated outflow concentrations presented in Figure 4. The overall upwards  
 338 trend of heavy metal concentrations (downwards for Mn), the fluctuations and peaks were all  
 339 reasonably modelled, with the Nash-Sutcliffe coefficient (NSE) values of 0.53 – 0.68 for PA and 0.56  
 340 – 0.64 for HP, respectively, indicating that the model can estimate the release of heavy metals from  
 341 two porous pavements with satisfactory; Mn was again an exception and had the poorest model  
 342 efficiency (E=0.13 for HP and 0.00 for PA), due to its complex potential removal processes and  
 343 variable performance observed.

344 The calibrated values of adsorption rate ( $k_{ads}$ ) and desorption rate ( $k_{des}$ ) had good agreement with the  
 345 pollutant removal performance observed, with higher  $k_{ads}$  and lower  $k_{des}$  values indicating relatively  
 346 higher removal rates; *e.g.* the particulate heavy metals (Pb, Al, and Fe) are characterized as higher  $k_{ads}$   
 347 and lower  $k_{des}$  values, vice versa for the soluble ones (Zn, Cu and Mn).

348 **Table 4** Performance of the model and calibrated parameters

	Porous Asphalt (PA)						Hydrapave (HP)					
	Calibration			Prediction			Calibration			Prediction		
	$k_{ads}$ (-)	$k_{des}$ (1/L)	$k$ (day/L)	NSE	Max NSE*	Obs. within prediction band (%)	$k_{ads}$ (-)	$k_{des}$ (1/L)	$K$ (day/L)	NSE	Max NSE	Obs. within prediction band (%)
Al	0.892	0.041	29.5	<b>0.63</b>	0.44	64	0.931	0.024	35.1	<b>0.56</b>	0.74	82
Cu	0.831	0.061	34.5	<b>0.59</b>	0.48	91	0.896	0.058	38.4	<b>0.62</b>	0.62	50
Fe	0.919	0.076	34.6	<b>0.56</b>	0.42	27	0.944	0.045	39.1	<b>0.53</b>	0.69	46
Pb	0.986	0.016	25.8	<b>0.64</b>	0.20	77	0.974	0.005	44.3	<b>0.53</b>	0.39	27
Zn	0.873	0.158	26.3	<b>0.64</b>	-0.62	46	0.952	0.126	44.2	<b>0.68</b>	0.04	22
Mn	0.455	0.214	5.21	<b>0.00</b>	-3.3	46	0.432	0.001	1288.1	<b>0.13</b>	-4.1	9

349 \*max NSE – the best of 100 model runs during model validation (*i.e.* prediction)

350 Figure 4 illustrates the 90% prediction bands as well as the best fit of prediction. Best performance  
351 was observed for Al and Cu in both systems, with 50%-91% of observations covered by the prediction  
352 bands (Table 4), and the trends and fluctuations were simulated reasonably well (max NSE = 0.44-  
353 0.74). Fe concentrations was over-predicted by the model, and the best fit of prediction in PA system  
354 extended slightly outside of the lower bound of the 90% prediction band; nevertheless, the overall  
355 prediction performance for Fe was acceptable, evidenced by the NSE of 0.42 for PA and 0.69 for HP.  
356 As for Pb, the peak concentrations after Year 15 were under predicted by the model, with relatively  
357 poorer model performance (77% and 27% observations within prediction band, max NSE of 0.20 and  
358 0.39 for PA and HP respectively). The model did not predict Zn concentrations well (NSE<0.04),  
359 especially for that in HP systems (best fit of the predictions were out of the 90% prediction bands and  
360 only 22% observations within the band), indicating that other important removal processes of these  
361 heavy metals not considered in the model (*e.g.* complexation with other compounds or biological  
362 transformation) might occur. Although the model predicted same decreasing trend of Mn  
363 concentrations as observed, it produced the worst model results, *e.g.* 46% and 9% for HP observations  
364 within prediction band, max NSE of <0 for PA and HP respectively. It should be acknowledged that  
365 90% prediction bands were generated in a strict way, *i.e.* using GLUE method based on top 1%  
366 parameter sets (corresponding to cut-off thresholds of NSE>0.45 for all the metals), which in some  
367 cases caused the best fit prediction fell out of the 90% prediction bands (*e.g.* Zn in HP system – it was  
368 checked that the whole prediction band overlapped with the best fit prediction); it however provides a  
369 stronger verification of the proposed model. Overall, the results indicate that model has abilities to  
370 predict long term performance of porous pavements for some heavy metals, *e.g.* most promisingly for  
371 Al and Cu with high NSE values and coverage of observations within the 90% bands, followed by Fe  
372 and Pb with lower coverage of observed data within the 90% prediction bands. The proposed model  
373 has to be improved further for predicting Zn and Mn removal by the porous pavements; *e.g.* it is  
374 suggested to include more removal processes of these two metals in model.

375 **Figure 4** Calibration and prediction results of outflow concentrations of the model. For PA, the 1-10  
376 year, and 1-13 year data were used for calibration of PA and HP respectively. The shaded areas  
377 indicate the 90% prediction band. Solid lines with symbols represent the best calibrated  
378 concentrations, while the solid lines without symbols represent predictions from max NSE.

379 The efficiency of the proposed model performance (NSE values equal to 0.44-0.74 for Al and Cu,  
380 0.42-0.69 for Fe and 0.20-0.29 for Pb) are reasonably good except for Zn and Mn (NSE from negative  
381 to 0.04) and have good agreements to previous reported water quality models for porous pavements;  
382 e.g. Newton (2005)'s one-parameter first decay model adapted from filtration theory can predict  
383 particle removal efficient from pavement systems under several individual events with different levels  
384 of satisfactory: low flow rates with NSE=0.36-0.98 and high flow rates with NSE from negative to  
385 0.39. He et al. (2015)'s empirical model developed based on laboratory data also provided good  
386 prediction on six field tests results for TSS and TP removal by porous concrete pavement with errors  
387 of up to 2.9% for average removal rates; the same study also tested a sediment removal conceptual  
388 model and reported prediction errors of 1.3% and 5.8% for TSS removal rates for two validation field  
389 events. As seen, these existing models are simple and can perform well; they however were just  
390 validated against individual events and could not be used for predicting long term performance of  
391 pavement systems that are exposed to continuous stormwater events. The current proposed model  
392 however has overcome these shortcomings and this study for the first time developed a new process-  
393 based model specifically for heavy metals involving both adsorption and desorption processes; more  
394 importantly its greater utility has been supported by the ability to simulate long-term treatment  
395 performance, thus can assist in better design of pavement systems.

#### 396 **4. Conclusions**

397 This study tested the long-term treatment performance of three porous pavements - Porous Asphalt  
398 (PA), Hydrapave (HP) and Permapave (PP) in removing heavy metals, using accelerated laboratory  
399 experiments spanning over 1 year simulating 26 years of operations. Intermittent dry/wetting periods  
400 were also simulated with varying flow conditions to represent a realistic operational scheme. A water

401 quality model that includes adsorption and desorption processes was proposed and tested using  
402 experimental data. The main findings are:

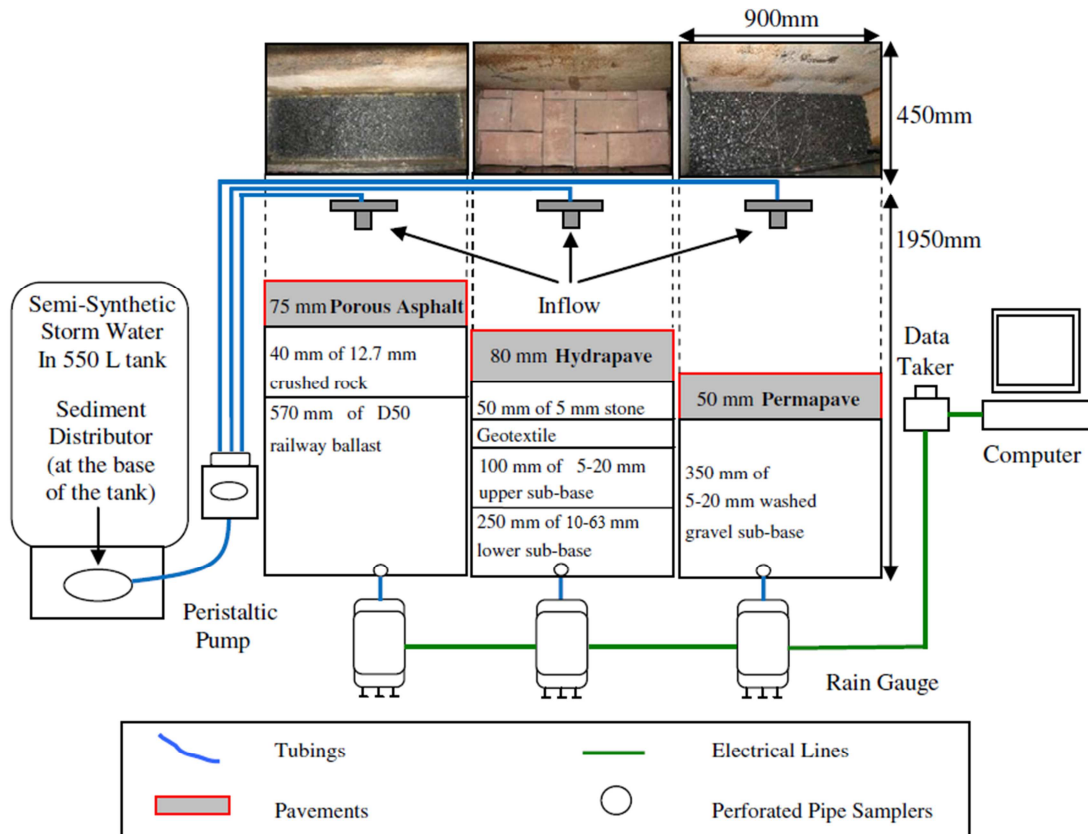
- 403 • All three porous pavements were good in removing heavy metals, with average cumulative  
404 removal rates over 26 years of operations being:  $84\pm 14\%$  for Pb,  $79\pm 13\%$  for Al,  $77\pm 13\%$  for  
405 Fe,  $68\pm 19\%$  for Fe,  $66\pm 20\%$  for Zn, and  $35\pm 35\%$  for Mn; metals with higher particulate  
406 fractions (Pb, Al and Fe) usually were easier to be retained by the pavements compared to  
407 soluble ones (*e.g.* Cu, Zn and Mn);
- 408 • Over the simulated 26-year period, effluent concentrations generally increased, indicating the  
409 long term release of heavy metals as a result of adsorption and desorption process; it was  
410 found that the porous systems took 1-2 years to mature for better and more stable  
411 performance. Clogging led to poorer system performance with higher variability.
- 412 • Permapave (PP) had the best performance as it was never clogged over the period, followed  
413 by Hydrapave (HP) and Porous Asphalt (PA); lower removals of soluble heavy metals was  
414 observed in the pavements with coarser sub-materials in PA;
- 415 • Flow rates influenced the heavy metal removal, with higher outflow concentrations usually  
416 corresponded to higher flowrate, especially for the soluble heavy metals; low flow rates was  
417 also not preferred especially after clogging occurred as it benefited to desorption processes  
418 and breakdown of accumulated sediments, leading to higher variability in heavy metal  
419 outflow concentrations.
- 420 • The proposed model was successfully calibrated against the data collected from first half of  
421 the accelerated experiment (*e.g.* 10~13 years), with the estimated NSE values 0.53 – 0.68  
422 (except for Mn which had NSE of 0.0-0.13);
- 423 • The prediction results indicate that the proposed model was promising for predict the  
424 releasing of Al and Cu from the porous pavements (50%-91% of observations covered by the  
425 prediction bands, max NSE = 0.44-0.74); it can also be applied for Fe and Pb, but with lower  
426 confidence (NSE= 0.42-0.69 for Fe and 0.20-0.29 for Pb) and smaller coverage of observed

427 data within the 90% prediction bands. The proposed model has to be improved further if it is  
 428 to be used for predicting Zn and Mn removal by the porous pavements.

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**Figure 1** The experimental set-up for testing Porous Asphalt, Hydrapave and Permapave (adapted from Yong et al. (2013))

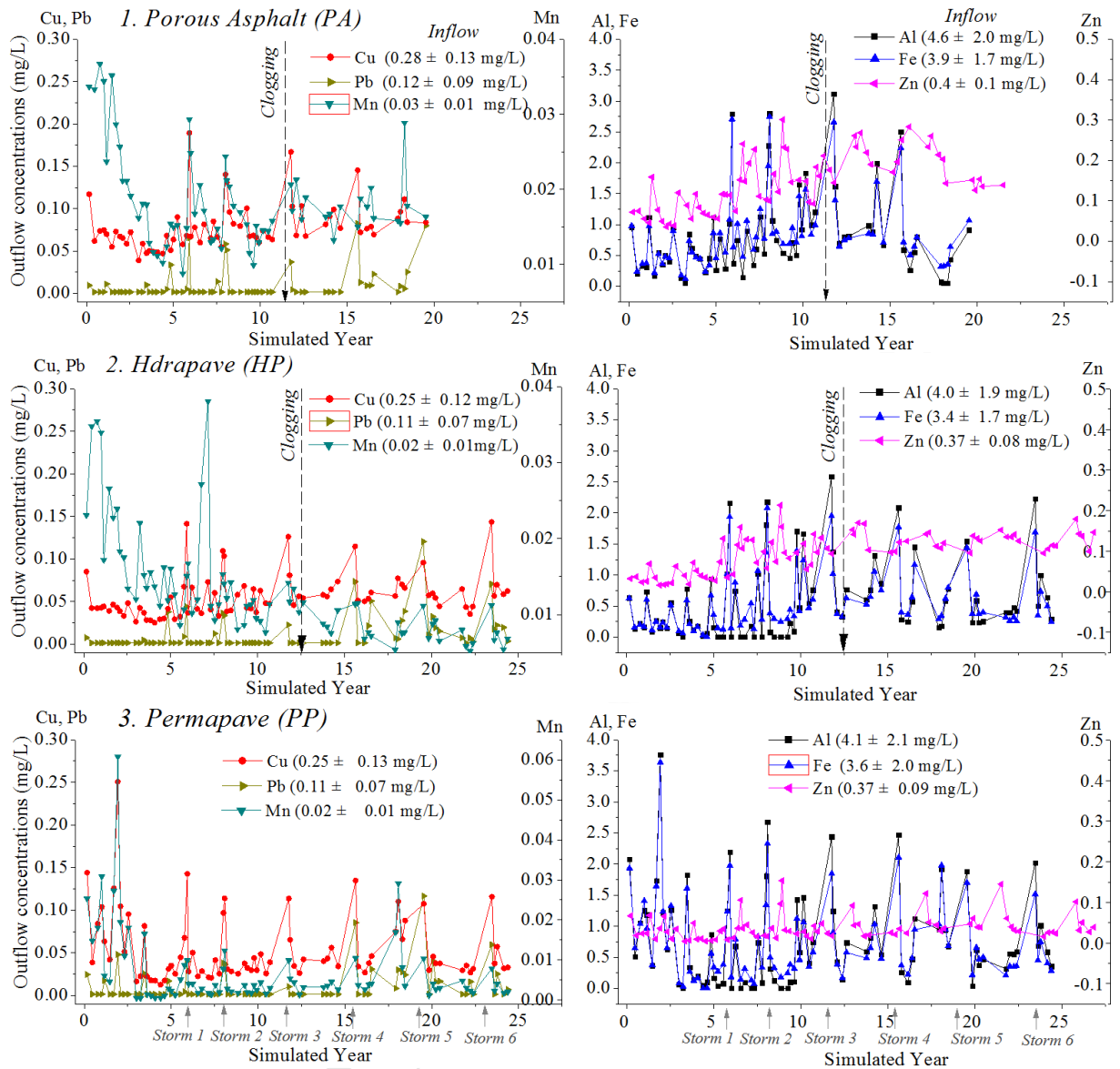


Figure 2 Change of outflow heavy metal concentrations over the course of 26 simulated years;

“average inflow concentration  $\pm$  standard deviation” indicated in brackets of the legends.



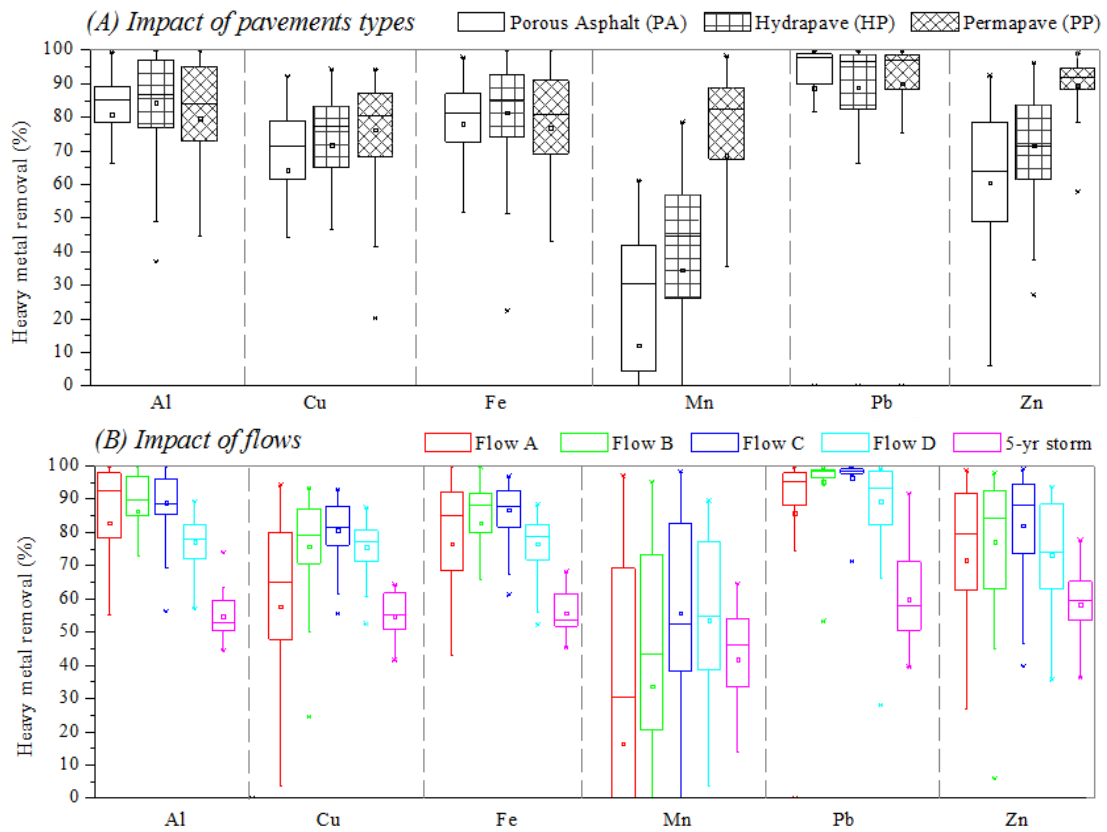
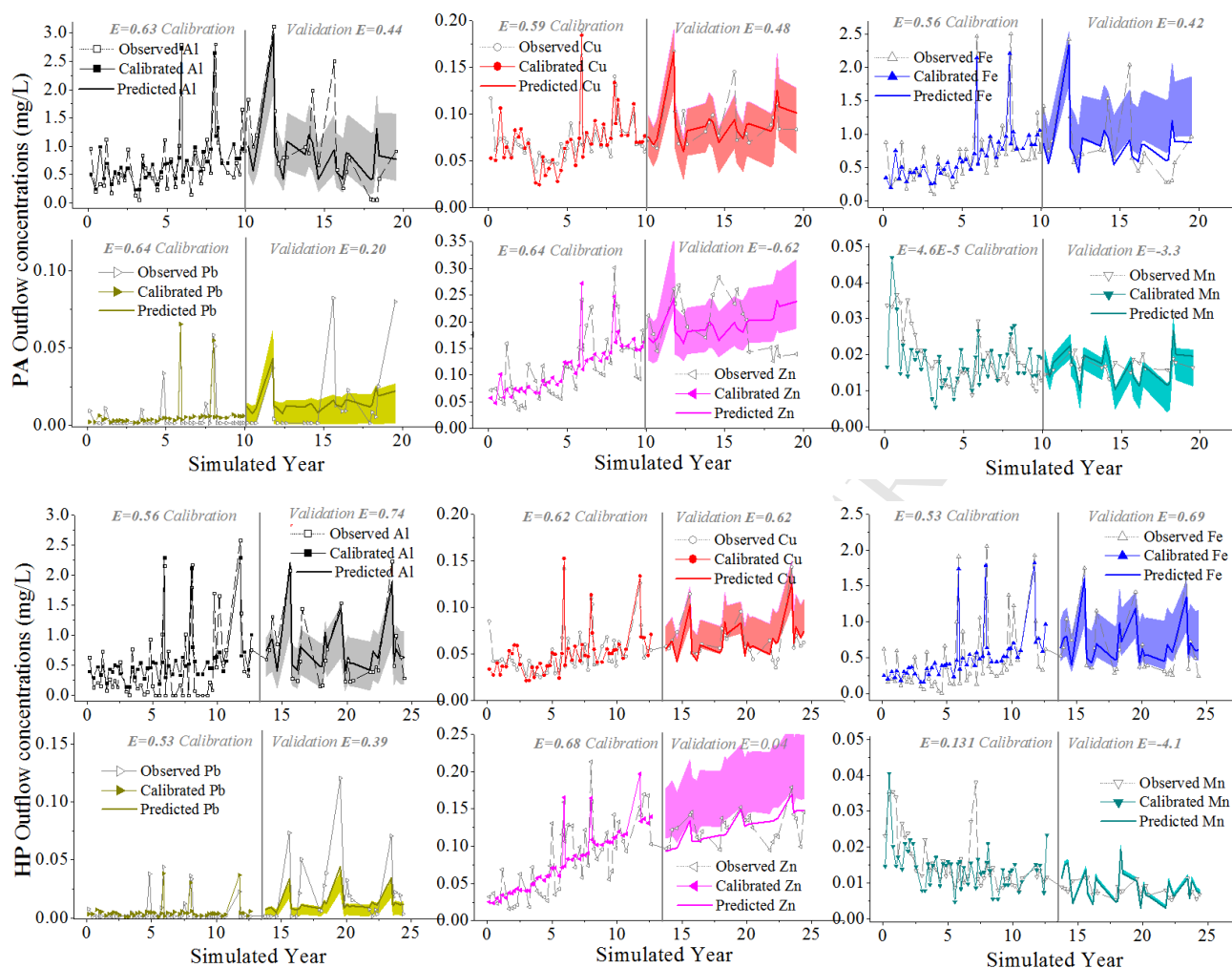


Figure 3 Influence of (A) pavement types and (B) flow rates on heavy metal removal



**Figure 4** Calibration and prediction results of outflow concentrations of the model. For PA, the 1-10 year, and 1-13 year data were used for calibration of PA and HP respectively. The shaded areas indicate the 90% prediction band. Solid lines with symbols represent the best calibrated concentrations, while the solid lines without symbols represent predictions from max NSE.

**Highlights**

- Long term metal removal by porous pavement was studied at varying conditions
- An increasing trend of outflow concentrations from was observed except for Mn
- Clogging led to poorer system performance with higher variability
- The first processed-based model was developed to predict heavy metal performance
- The model was promising in predicting Al and Cu removal, followed by Fe and Pb