

Mesoporous silica nanoparticles in target drug delivery system: A review

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Abstract

Due to lack of specification and solubility of drug molecules, patients have to take high doses of the drug to achieve the desired therapeutic effects for the treatment of diseases. To solve these problems, there are various drug carriers present in the pharmaceuticals, which can be used to deliver therapeutic agents to the target site in the body. Mesoporous silica materials become known as a promising candidate that can overcome above problems and produce effects in a controllable and sustainable manner. In particular, mesoporous silica nanoparticles (MSNs) are widely used as a delivery reagent because silica possesses favorable chemical properties, thermal stability, and biocompatibility. The unique mesoporous structure of silica facilitates effective loading of drugs and their subsequent controlled release of the target site. The properties of mesoporous, including pore size, high drug loading, and porosity as well as the surface properties, can be altered depending on additives used to prepare MSNs. Active surface enables functionalization to changed surface properties and link therapeutic molecules. They are used as widely in the field of diagnosis, target drug delivery, bio-sensing, cellular uptake, etc., in the bio-medical field. This review aims to present the state of knowledge of silica containing mesoporous nanoparticles and specific application in various biomedical fields.

Key words: Bio-distribution, evaporation-induced self-assembly, mesoporous silica nanoparticles, silica precursors, surface functionalization, target drug delivery

INTRODUCTION

The era of nanotechnology has revolutionized the drug delivery and targeting process and changed the landscape of the pharmaceutical industry. Nanoparticles have dimension below 0.1 μm or 100 nm especially in the drug delivery. The drug is dissolved, entrapped, encapsulated or attached to a nanoparticle matrix.^[1] Nanoparticles size range also affects the bioavailability and bio-distribution of particles, and hence it is useful as a drug carrier.^[2] Hydrophobic core is beneficial for drug loading while hydrophilic surface blocks opsonization and allow easier movement in the system. Numerous nanodevices have been reported like carbon nanotubes, quantum

dots, and polymeric micelles, etc., in the field of nanotechnology. In the present scenario, mesoporous nanoparticles are emerging for their well-known drug delivery and targeting purposes. Historically, Kresge *et al.* have described a means of combining sol-gel chemistry with liquid-crystal-line templating to construct ordered porous molecular sieves characterized by periodic arrangements of uniformly sized mesopores (diameter between 2 nm and 50 nm) incorporated within an amorphous silica matrix.^[3]

Mesoporous silica nanoparticles (MSNs) have become apparent as a promising and novel drug vehicle due to their unique mesoporous structure that preserving a level of chemical stability, surface functionality and biocompatibility ensure the controlled release, and target drug delivery of a variety of drug molecules.^[3,4] Mesoporous silica materials were discovered in 1992 by the Mobil Oil Corporation have received considerable attention due to their superior textural properties such as high surface area, large pore volume, tunable pore diameter, and narrow pore size distribution.^[5]

Mesoporous nanoparticles have a solid framework with porous structure and large surface area, which allows the attachment of different functional groups for targeted drug moiety to a particular site.^[6] Chemically, MSNs have honeycomb-like structure and active surface. Active surface enables functionalization to modify surface properties and link therapeutic molecules.^[7] Various pore geometric structures were shown in Figure 1.

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Mesoporous silica nanoparticles due to their low toxicity and high drug loading capacity, so they are used in controlled and target drug delivery system. Basically, silica is widely present in the environment in comparison to other metal oxides like titanium and iron oxides it has comparatively better biocompatibility.^[8] The mesoporous form of silica has unique properties, particularly in loading of therapeutic agents at high quantities, and in the subsequent releases. Due to strong Si-O bond, silica-based mesoporous nanoparticles are more stable to external response such as degradation and mechanical stress as compared to niosomes, liposomes, and dendrimers which inhibit the need of any external stabilization in the synthesis of MSNs.^[3,9] The mesoporous structure such as pore size and porosity can be tuned to the size and type of drugs. Various types of MSNs with internal structure and pore size as shown in Table 1. Another distinctive advantage of MSNs is that they have well-defined surface properties that allow easy functionalization of the silanol-containing surface to control drug loading and release.^[10,11] The surface functionalization is generally needed to load proper type of drug molecules (hydrophobic/hydrophilic or positive/negative charged), specific actions can also have a natural quality or characteristics by the functionalization through chemical links with other materials such

as stimuli-responsive, luminescent or capping materials, leading to smart, and multifunctional properties.^[12]

The mesoporous particle could be synthesized using a simple sol-gel method or a spray drying method. Mesoporous silica has been also widely used as a coating material. MSNs density can be increased by two strategies that is, capping of the pores of MSNs with gold nanoparticles and gold plating on MSNs surface.^[11]

Synthetic amorphous silica consists of nano-sized primary particles of nano or micrometer-sized aggregates and of agglomerates in the micrometer-size range. It has been used in a wide variety of industrial and consumer applications including food, cosmetics, and pharmaceutical products for many decades. Based on extensive physico-chemical, ecotoxicology, toxicology, safety, and epidemiology data, no environmental or health risks have been associated with these materials. It does not produce any toxicological effects on medicines and approved by generally recognized as safe.^[13]

Nanoparticles especially which are made from a pyrolysis method for instance, can be coated with a layer of silica and produce good stability in aqueous solutions. Both applications can benefit from the facile surface chemistry of silica that allows easy coupling of targeting ligands onto the particles.^[4,6] MSNs have been used in controllable drug delivery, gene transport, gene expression, biomarking, biosignal probing, imaging agent, detecting agent, drug delivery vehicles, and other important biological applications

Conventional MSNs can load a dose of therapeutic drug with 200-300 mg (maximally about 600 mg) drug/1 g silica. However, hollow MSNs also called MSNs with hollow core-mesoporous shell structure are able to achieve a super-high drug loading capacity because it provides more space to load drugs due to the hollow cores, typically >1 g drug/1 g of silica.^[18]

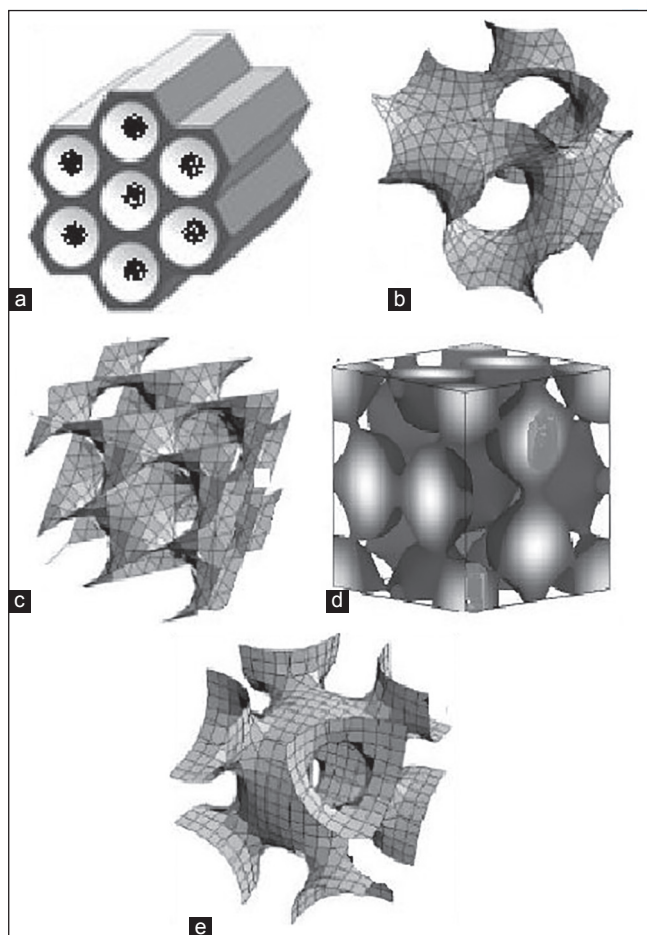


Figure 1: Various pore geometrics of mesoporous structure (a) 2D hexagonal p6 mm, (b) bicontinuous cubic Ia3d, (c) bicontinuous cubic pn3 m, (d) cage type pm3n, (e) cage type Im3 m

SYNTHESIS OF MESOPOROUS SILICA NANOPARTICLES

The synthesis of MSNs occurs at a low surfactant concentration to make the structure of the ordered mesophases strongly dependent on the interaction between the growing anionic oligomers of

Table 1: Various types of mesoporous silica nanoparticles with their internal structure and pore diameter

Type	Internal structure	Pore diameter (nm)	References
MCM-41	2D hexagonal	1.5-3.5	[14]
MCM-41	Hexagonal structure with uni-directional pore structure	3.70	[15]
SBA-15	2D hexagonal	6.0-10.0	[14]
SBA-15	2D hexagonal	7.80	[15]
SBA-15	3D cubic cage like	4.0-9.0	[16]
MCM-48	3D cubic	2.5-3.0	[17]

MCM: Mobile crystalline material, SBA: Santa barbara amorphous — type mesoporous

orthosilicic acid and cationic surfactant, which changes limits of the structure of mesophases to small sizes.^[3,18]

Mesoporous silica nanoparticles synthesis based on solution

The most widely used types of MSNs are mobile crystalline material (MCM-41), it consists of ordering hexagonally arrangement of cylindrical mesopores. Synthesis of MCM-41 required liquid crystal is templating of an alkyl ammonium salt that is, cetyl trimethyl ammonium bromide.^[19] High concentration of amphiphilic surfactant assembles into a spherical micelle in the water and hydrophilic soluble precursor like polysilicic acid or silica acid. By electrostatic and hydrogen bonding interaction, the silica precursor is concentrated at the hydrophilic interface and form an amorphous silica, which is a mold of the mesoporous product. Removal of remaining surfactant can be done by calcination and extraction method.^[20]

Evaporation-induced self-assembly

This method was established in 1997. It is starting by forming a homogeneous solution of soluble silica and surfactant in ethanol, water with an initial surfactant concentration of critical micelle concentration. The Solvent evaporation process will start during dip coating for increase surfactant concentration.^[21] Then driving a mixture of silica/surfactant micelles, and their further formation occur into liquid crystalline mesophases as shown in Figure 2. Film process was done by use of aerosol processing to direct the formation of mesoporous nanoparticles.^[18,21] Evaporation-induced self-assembly is a non-volatile component that can be introduced into an aerosol droplet incorporated within the MSNs.

SOL-GEL PROCESS

This process requires two-step consideration: Hydrolysis and condensation. Hydrolysis produced colloidal particles in aqueous solution, which can be stimulated at alkaline and acidic pH.^[22] At neutral pH, condensation reaction takes place in which gel-like 3D network structure formed by cross-linking through siloxane bond. After drying at ambient temperature the different biomolecules embedded in a matrix structure of silica gel as shown in Figure 3. This process involves the formation of the (MCM-41) under the

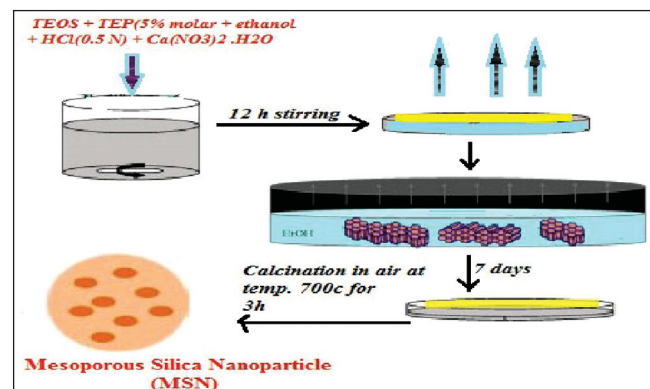


Figure 2: Evaporation self-induced assembly

size range of 60-1000 nm. Several advantages of the sol-gel process like, it is a simple and cost effective process used to provide MSNs with controlled mesoporous structure and surface properties.^[23] Sol-gel process is not a multistep process, so its time-saving process and required less excipients

Synthesis mechanism of mesoporous silica nanoparticles

The synthesis process of MSNs involves the replication of a surfactant liquid crystal structure and polymerization of metal oxide precursor. Removal of organic surfactant through the calcination process which form a porous structure. Various chemical constituents used in the formation MSNs were described in [Table 2].

Silica Precursor used in mesoporous silica nanoparticles.

Organically modified precursors

They are prevented hydrolysis because an organic group attached directly to a silicon atom, which does not need oxygen bridge. It is conceded that organo-silica nanoparticles consist of better properties including large surface area, less condensed siloxane structure, and low density.^[24] The limited accessibility and high cost of organic template lead to its restricted use in practical applications. Commonly used silica precursors are glycerol-derived polyol-based silanes, orthosilicic acid, sodium metasilicate, tetraethyl orthosilicate (TEOS) or tetramethoxysilane (TMOS), and tetrakis (2-hydroxyethyl) orthosilicate.^[3,25]

1. Tetraethyl orthosilicate or TMOS was commonly used in MSNs synthesis. However, their poor water solubility requires additional organic solvent and alcohol and needs extreme conditions of pH and high temperature, which restricts their use.^[26,27]
2. Tetrakis (2-hydroxyethyl) orthosilicate had been investigated to address the problems associated with TEOS and TMOS. It is now used in many studies as MSNs precursor because it is more biocompatible with biopolymers, more water soluble

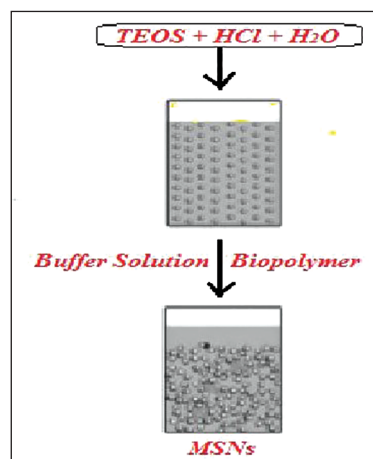


Figure 3: Sol-gel process in the synthesis of mesoporous silica nanoparticles

than TEOS and TMOS, and can process jellification at ambient temperature with a catalyst.^[27]

Glycerol-derived polyol-based silane precursors

They are not pH dependent but very sensitive to the ionic strength of the sol. This can form optically clear monolithic MSNs. The residuals can be either removed or retained, therefore, the shrinkage during long-term storage can be minimized.^[28] Orthosilicic acid was used as a silica precursor in the past but due to the extensive time consumption and requirement of freshly prepared acid, so it is not widely used anymore now a day.^[29]

Sodium metasilicate

It is another precursor to sol-gel-derived silica. Formation of sodium chloride was investigated, which can cause a problem if a significant amount is generated. Later researches suggested that removing of this salt formulation by dialysis process, but it is a time and cost consuming procedure.^[30] Hence, alkoxides and pure alkoxysilanes are currently widely used.

ADVANTAGES AND DISADVANTAGES OF POROUS SILICA MATERIAL

Porous silica based materials are among the most beneficial compounds which can provide more opportunities for treatment of cancer therapy and provide a pathway toward the treatment of challenging diseases. Silica or silicon has various versatile and broad range advantages such as versatility, non-toxicity, biocompatibility, biodegradability, high surface area, pore volume, homogenous distribution of guest molecules in porous space, the ability for surface charge control, and free dispersion throughout the body.^[31] The major disadvantage of porous silica nanoparticles is attributed to the surface density of silanol groups interacting with the surface of the phospholipids of the red blood cell membranes resulting in hemolysis. Another disadvantage is related to metabolic changes induced by porous silica nanoparticles leading to melanoma promotion.

Biocompatibility and bio-distribution of mesoporous silica nanoparticles

The use of the organic compound into organism requires determination of biocompatibility and bio-distribution of MSNs. On the behalf of a study conducted by Hudson *et al.* suggested that intraperitoneal or intravenous produce a lethal effect in mice but subcutaneous injection showed no toxicity.^[23,32] On the basis of characteristics property, MSNs were targeted to different cell as describe in Figure 4. Hence, both factors depend upon high dose and large particles with consideration to surface properties, shape, and size of MSNs.

Biocompatibility is the study of the interaction of biomaterials with the body of experimental animals to evaluate possible toxicity effects.^[40]

Biocompatibility study in mice

A dose-escalating study of phosphonate-MSNs (100-130 nm) to examine biocompatibility of the material and it was shown that repeated doses of phosphonated MSNs ranging from 5 mg/kg to 50 mg/kg in mice did not result in significant toxicity as determined by body weight, histology, and hematological

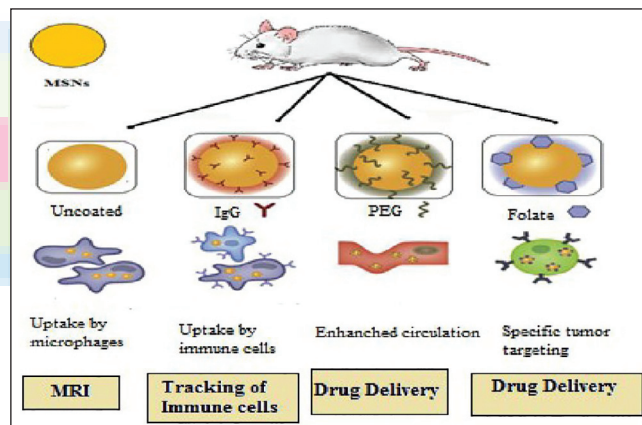


Figure 4: Biocompatibility and bio-distribution of mesoporous silica nanoparticles

Table 2: Common chemical constituents used in preparation of MSNs

Substrate	Function	References
N-dodecanonyl-β-alanine	Surfactant with an amino acid residue	[8]
CTAB	Increase water solubility of hydrophobic ligand	[9], [31]
PEG	Improve biocompatibility and functional characteristics of silica matrix	[32]
Tween 80	Surfactant	[8]
PVA	Settle down gel in THEOS-containing solution	[33]
PEO	Induce hydration PEO/sol ratio regulates size	[34], [35]
Sodium hydroxide	Catalyst	[32]
Hydrogen fluoride	Catalyst	[9], [31]
Hydrogen chloride	Catalyst	[32], [33], [36]
Nonionic triblock copolymer	Structure-directing agent	[9]
Trihydroxysilylpropylmethyl phosphate	Prevent-interplace aggregation	[31]
Ammonium nitrate	Surfactant removal	[31]
Methanol	Solvent in TMOS	[37], [38]
Ethanol	Solvent in TEOS	[38]

CTAB: N-cetyl trimethyl ammonium bromide, PEG: Polyethylene glycol, PVA: Polyvinyl alcohol, PEO: Polyethylene oxide, TMOS: Tetramethoxysilane, TEOS: Tetraethyl orthosilicate, MSN: Mesoporous silica nanoparticles, THEOS: Tetrakis (2-hydroxyethyl) orthosilicate, SBA: Santa barbara amorphous

studies.^[41] A short term treatment of five injections in 14 days and a long-term study treatment of 18 injections were carried out in mice produce no major side-effect. Different surface modification and particle size have on biocompatibility in nanoparticles, so these modifications can affect the interaction of nanoparticles in different organs and tissue.^[28,30]

Bio-distribution is defined as tracking the localization of compounds of interest in the organs and tissues of experimental animals to determine the accumulation in different parts of the body. The biodistribution process using MSNs was shown in [Figure 5].

Bio-distribution has also been study showed that each type of porous silica (PSi)-particles tend to be removed from the body via the complement system or urine depending on the administration route. Bio-distribution of MSNs accumulated in tumors at concentrations ranging from 45 ng/mg (4 h postinjection) to 110 ng/mg (24 h postinjection) before decreasing to 65 ng/mg (48 h postinjection).^[42] The highest accumulation of nanoparticles was found to be in the kidney and lungs. The other organs, including liver, heart, intestine, and spleen had a low accumulation in 48 h postinjection ranging from 3 ng/mg to 10 ng/mg. Targeting of the MSNs of cancer cells by attaching folate to their surface resulted in an increase in tumor accumulation. Experimental study showed that positively charged nanoparticles are excreted from the liver into the gastrointestinal tract and excreted from the animal in the feces while negatively charged MSNs are sequestered within the liver.^[43] Therefore, the charge on the surface of the nanoparticles may play an important role in their bio-distribution. Another property that affects the bio-distribution of material used is the shape of the nanoparticles. Huang *et al.* showed that MSNs with a short rod-shape accumulate in the liver and long rod-shaped nanoparticles accumulate in the spleen.^[42] The excretion of MSNs through the urine and feces has been reported by various groups. This study using phosphonate-MSNs up to 94% of the silica was excreted from the mice 4 days after injection with 73% being detected in the urine and 21% in the feces

Functionalization of mesoporous silica nanoparticles

Co-condensation process allows direct functionalization of the silica surface during the synthesis of the material, which was shown to be preferential for the internal surface of the mesopores.

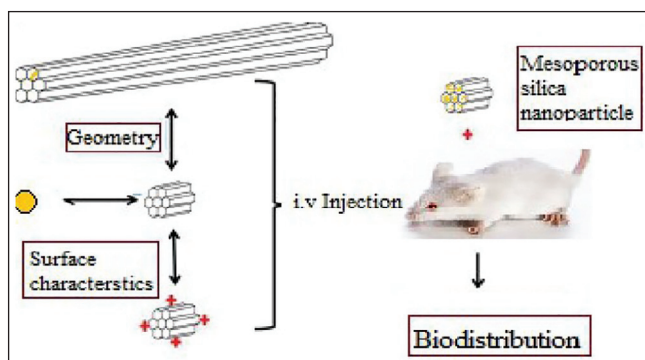


Figure 5: Biodistribution process of mesoporous silica nanoparticles

Typically, organosilanes (R-TMS or R-TES) where R is an organic functional group, while TMS is trimethoxysilane and TES is triethoxysilane is added into the reaction solution together with TEOS or immediately after the addition of TEOS which affords incorporation of organosilanes in the final material.^[43]

Postsynthesis strategies for functionalization

Postsynthetic grafting strategy

In general, postsynthesis grafting is useful for functionalizing mesoporous silicates with active molecules that are incompatible with the chemistry of the initial sol. It is often desirable to functionalize accessible silica surfaces, such as the surface of nanoparticles or areas near pore orifices.^[44] Postsynthesis grafting involves the same types of silane linkers that are commonly utilized in the bifunctional strategy.

Backfilling strategy

Backfilling is a simple strategy for introducing active molecules into the empty pores of mesostructured silica. Empty mesopores are filled by exposing a material to a solution or vapor containing the active molecules and allowing them to diffuse into it.^[45] Backfilling does not result in chemically modified materials but is less widely useful.

Surface modification

Various characteristics of MSNs like the Surface area, zeta-potential, hydrophilicity, functional groups on surface, affinity, and selectivity that influence the bioavailability of the particles. By changing these characteristic properties of the particle modified more effective drug carriers.^[46]

Target specificity

Mesoporous silica nanoparticles have ability to reach the correct target tissue mainly tumor on the basis of surface modification. Selectivity of nanoparticles improved by conjugation various target molecules via covalent bonding. Widely used targeting agents are organic molecules that already exist in the system as they usually are nontoxic, biodegradable and stable in the wide range of pH.^[45]

Polyethylene glycol

Polyethylene glycol (PEG) is also nontoxic and causes no immunological response improving further its usefulness in surface modifications of nanoparticles. A surface modification of nanoparticles is PEGylation, in which the particles are coated with PEG chains. PEG can be attached to the particles with either covalent bonds, adsorption or by entrapping PEG moieties within the surface.^[47,48]

APPLICATIONS OF MESOPOROUS SILICA NANOPARTICLES

MSNs shows various applications in the different field as shown in Figure 6.

Drug delivery

Porous silica based materials for drug delivery applications emerged in the late 1990's with reports of silica nanopore membranes; one major advantage of the PSi materials is that they can be tailored for continuous or triggered drug release depending on the application.^[30] The paracellular delivery of insulin across an intestinal Caco-2 cell monolayer using micro fabricated PSi particles was one of the first examples of drug delivery using this material. Because of high surface area, the selectivity of the adsorption substance and low toxicity MSNs were widely used in adsorption of toxic molecules and drug delivery system.^[49,50] In general, people use activated carbon to remove toxic chemicals however it showed poor removal ability

Conventional MSNs can load a dose of therapeutic drug with 200-300 mg (maximally about 600 mg) drug/1 g silica. For example, conventional glucose-responsive insulin delivery systems suffer from the decrease of insulin release with repeated cycles^[18,51] [Table 3].

Cellular uptake of mesoporous silica nanoparticles

Silica based material nanoparticles uptake by cell produce no detectable toxic effects.^[58] The mechanism of cellular uptake appears to be mediated by an active endocytosis pathway as MSNs endocytosis is inhibited by a decrease in temperature to 4°C that

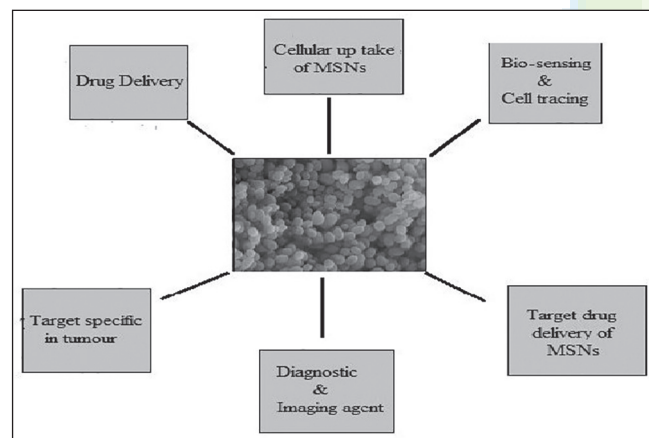


Figure 6: Applications of mesoporous silica nanoparticles in various bio-medical fields

is incubation with metabolic inhibitors. Functionalization of the external surface of MSNs with groups in which cells express specific receptors like folic acid, which notably enhances the uptake efficiency of the material by cells and functionalization of the particles with groups that alter their zeta-potentials affects not only the efficiency of their internalization, but also the uptake mechanism.^[56,59]

Bio-sensing and cell tracing

Mesoporous silica nanoparticles used as bio-sensing element due to their size and versatile chemistry of the structure. The unique surface properties and the small particle sizes of these nanoparticle-based sensor systems used for the detection of analysts within individual cells both *in vivo* and *in vitro*.^[60] Nanoparticles do not suffer from fluorescent, self-quenching, and other diffusion-related problems. Therefore, the ability to functionalize the surface of nanoparticles with large quantities of cell-recognition or other site-directing compounds produces as an ideal agent for cell tracing.^[42,49]

Target specific in tumor

The application of porous silica materials for cancer therapy has been emerging as a new interesting field of interdisciplinary research among chemistry, medicine, material science, biology, and pharmacology.^[22,34] Selective targeting strategies employ ligands that specifically bind to receptors on the cell surface of interest to promote nanocarrier binding and internalization. This strategy requires that receptors are highly overexpressed by cancer cells relative to normal cells.^[53] The effective drug treatment by MSNs in the cancerous cell as shown in Figure 7. The outstanding advantages of porous nanomaterials is the ability of surface functionalization with targeting moieties is the most exciting way to deliver the drug to a cancerous cell.

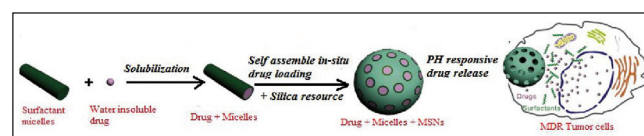


Figure 7: Present multi-drug delivery systems of mesoporous silica nanoparticles in targeted delivery in the treatment of tumor cells

Table 3: List of various types of drugs delivered through mesoporous silica nanoparticles

Classes	Drugs	Delivery medium	References
Anti-inflammatory	Ibuprofen	SBA-15	[16], [52]
	Naproxen	SBA-15	[52]
	Naproxen	MCM-4	[52]
Chemotherapy	Naproxen	Amine-modified MCM-41	[52]
	DOX	Folic-acid conjugate MSN	[53]
	DOX	DOX-hydrozone MSN-FA	[32]
Estrogenic	Camptothecin	Galactose-functionalized MSN	[32]
	Alendronate	SBA-15 with phosphorous	[54], [55]
	Alendronate	Amine-modified MCM-41	[54], [55]
Antibiotics	Alendronate	MCM-41	[56], [57]
	Erythromycin	Octadecyl-functionalized SBA-15	[16]
	Erythromycin	SBA-15	[16]
	Vancomycin	CdS-capped MCM-41	[57]

MCM: Mobile crystalline material, MSN: Mesoporous silica nanoparticles, SBA: Santa barbara amorphous, DOX: Doxorubicin, FA: Folic acid

Targeted MSNs therapies are used to block the growth and spread of cancer by interfering directly with specific molecules involved in tumor growth and progression or indirectly by stimulating the immune system to recognize and destroy cancer cells.^[61] For example, the delivery of many toxic antitumor drugs requires “zero release” before reaching the targeted cells or tissues. Efficient delivery of doxorubicin (DOX) using MSNs coated with a PEG copolymer using 50 nm MSNs, when coated with the copolymer they can reach a size of 110 nm.^[16,62] Cancerous therapy was done through injected of 120 mg/kg of nanoparticles on a weekly basis for 3 weeks to a KB-31 xenograft model. The DOX-loaded nanoparticles demonstrated 85% tumor inhibition compared with 70% inhibition by the free drug.^[63] Therefore, there is an important need for using porous nanoparticles as nanovaccines to treat cancer

Diagnostic and imaging agent

In recent years, multifunctional nanomaterials are used for simultaneous imaging and therapy. The field has expanded as rapidly as “theranostic” has been coined to describe platforms that serve dual roles in the field of diagnostic and therapeutic agents.^[64,65] Ideally, these nanomaterials will be suitable for long-term quantitative imaging at low doses and safely cleared from the body after imaging is complete. Silica-based imaging nanoprobe are most commonly used for optical, magnetic resonance imaging (MRI) or a combination of both modalities.^[66] Silica nanoparticles are an excellent carrier for facile loading of a wide variety of imaging and therapeutic moieties, making them promising candidates for theranostic applications

Targeted drug delivery of mesoporous silica nanoparticles

Specific targeting is highly attractive as an approach to spontaneously distinguish the site of disease diagnosis and as a result, this technique reduces drug administration dosage and diminishes toxic side-effects of drugs during circulation.^[67] Both passive strategies and active surface decoration methods have been applied to the fabrication of novel MSNs based drug delivery systems for targeted release.^[68]

Passive strategies

Passive routes passive accumulation of MSNs in tumor tissue can be realized by the enhanced permeability and retention (EPR) effect, a theory first postulated by Matsumura and Maeda.^[69] They hypothesized that the differential localization of macromolecules, as well as particles of certain sizes, is attributed to the tumor

microenvironment, the relative slow elimination rate, and poor lymphatic drainage. Effectiveness of the EPR effect can be mediated by the particle size, surface charge or hydrophobicity.^[70,71]

Active surface decoration

Surface decoration with targeting ligands efforts have been made to functionalize the surfaces of MSNs with cancer-specific targeting ligands for an enhanced MSNs uptake by cancer cells compared to non-cancerous cells. One such ligand is folic acid,^[72] as folate receptors are known to be overexpressed in several types of human cancer including ovarian, endometrial, colorectal, breast, and lung.^[73] Besides folic acid, other small cell nutrient molecules such as mannose and 100 were also shown to selectively improve the uptake of MSNs by breast cancer cells.

PATENTED CITATION OF MESOPOROUS SILICA NANOPARTICLES

The various types of patent in the field of MSN were shown in [Table 4].

Future prospective

Mesoporous silica nanoparticles involve the identification of precise targets (cells and receptors) related to specific clinical conditions, and it is a choice of the appropriate nanocarriers to achieve the required responses at target site while minimizing the side effects of respective target molecule. Thus, *in vivo* potential of nanotechnology toward targeted imaging and drug delivery to be realized, that MSNs have to get smarter effects on a wide range of treatment in various diseases. MSNs are targeted contrast agents that improve the resolution of cancer cells to the single cell level. It is act as good diagnostic agents if iron oxide nanocrystals where incorporated into the mesoporous particles for magnetic manipulation and MRI. To further develop mesoporous silica for oral administration, the mechanisms behind the effects on weight and fat reduction needs to be investigated fully. Therefore, it is essential that fundamental research be carried out to address these issues if the successful efficient application of these technologies is going to be achieved.

CONCLUSIONS

In conclusion, this review describes in detail the recent progress of synthesizing and functionalizing MSNs and using these nanomaterials as targeted release and biological cell delivery

Table 4: Patented citations

Cited patent	Topic	References
US5639603	Synthesizing and screening molecular diversity	[74]
US5981180	Multiplexed analysis of clinical specimens apparatus and methods	[75]
US6180415	Plasmon resonant particles, methods, and apparatus	[76]
WO 2004063387 A2	Fluorescent silica-based nanoparticles	[77]
20120207795	cationic polymer coated mesoporous silica nanoparticles and uses	[78]
US20030152759	Method of fabricating nanostructured materials	[79]
US20060018966	Antimicrobial mesoporous silica nanoparticles	[80]
US20070275068	Controlled release delivery system for bio-active agents	[81]

vehicles. On the other side, the synthesis methods to functional MSNs with a core-shell structure are able to produce an acceptable amount of nanoparticles. Hence, they can be used as a unique essential drug carrier to overcome unwanted side effects. A versatile approach for the selective functionalization of MSNs has been developed. Hence, they have a wonderful approach for the application in the different field like drug delivery, diagnostic and imaging agent, and target drug delivery in the treatment in cancer. It is important to underline that one of the most valuable characteristics of silica nanoparticles is the possibility to use them to merge different materials, to combine various functionalities, that is to say to act as a basis to obtain multifunctional medicine dedicated nanoplatforms enabling multimodal imaging and simultaneous diagnosis and therapy. MSNPs are promising nanocarriers to efficiently transport and site-specifically deliver highly toxic drugs, like chemotherapeutic agents for cancer treatment. MSNs are efficiently transported site specific drug delivery of highly toxic drug such chemotherapeutic agents for effective cancer treatment. They have stimuli responsive drug releases so, it enhancing and minimizing the side effects of anti-cancer drugs in cancer therapy.

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