

High hydrostatic pressure as emergent technology for the elimination of foodborne viruses

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High hydrostatic pressure is a non-thermal technology that eliminates microorganisms with a milder effect on the quality of the foods than that produced by heat treatment. Consequently it can produce microbiologically safe foods, with an extended commercial shelf life and with better characteristics compared to heat-treated foods. Whereas the effect of this technology on foodborne pathogenic bacteria has been extensively studied, there is less information on pressure inactivation of enteric viruses. In this article, we review recent studies on the elimination of foodborne viral risks, and detail the different parameters which could influence the inactivation.

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Use of HHP in food industry

High hydrostatic pressure (HHP) is a non-thermal process that inactivates pathogenic and spoilage microorganisms as well as endogenous enzymes, preserving the sensorial characteristics and prolonging the shelf life of food products (Considine, Nelly, Fitzgerald, Hill, & Sleator, 2008; Patterson, 2005). However, the response of different types of microorganisms varies significantly; i.e. vegetative bacteria are the most sensitive group to HHP, followed by yeasts and moulds, while viruses and bacterial spores are the most resistant (Patterson, Linton, & Doona, 2007).

HHP is being used in recent years in the food industry as an alternative to a wide range of food processing technologies, especially to thermal processing (Buckow & Heinz, 2008; Considine *et al.*, 2008; Fonberg-Broczek *et al.*, 1999, 2005). It is an energy-efficient and rapid process that can allow short processing times (Buckow & Heinz, 2008; Farr, 1990; Knorr, 1995; Patterson *et al.*, 2007). It uses pressure of up to 1000 MPa; this is transmitted isostatically and instantaneously, and thus the process is independent on the shape or size of the food, which can often be problematic in thermal processing of large food items (Farr, 1990; Knorr, 1999; Smelt, 1998). Compared to thermal processing, pressure has less detrimental effects on food and therefore the products preserve most of their natural colours and flavours and health-promoting substances (Kingsley, Guan, & Hoover, 2005; Wilkinson, Kurdziel, Langton, Needs, & Cook, 2001).

The first HHP-treated product that appeared on the market was a high acid jam in Japan in the early 1990s. Since then, a wide spectrum of food products have been commercialised, for example fish and seafood products, meat products such as cook or cured ham, fruit products such as guacamole, fruit jellies and juices, and ready-to-eat (RTE) products (Considine *et al.*, 2008; Goh, Hocking, Stewart, Buckle, & Fleet, 2007; Murchie *et al.*, 2005; Smelt, 1998; Torres & Velazquez, 2005).

Effect of HHP treatment on food

HHP can help to maintain the quality attributes of fresh food, rendering products microbiologically safe with an extended shelf life (Hogan, Kelly, & Sun, 2005; Patterson, 2005). However HHP can sometimes affect the food yield, sensory qualities such as colour and texture, and produce biochemical changes affecting negatively to the food

properties, but these effects are less severe than those experienced using thermal processing techniques (Buckow & Heinz, 2008; Hogan *et al.*, 2005). In addition, those side effects on food properties can be attenuated by a suitable selection of the processing parameters: temperature, time and pressure (Buckow & Heinz, 2008).

HHP, in contrast to heat, does not disrupt covalent bonds thus maintaining the primary structure of proteins, but does alter the conformation of proteins by causing irreversible changes to the secondary, tertiary, quaternary, and supramolecular structure (Murchie *et al.*, 2005; Palou, Lopez-Malo, Barbosa-Canovas, & Swanson, 1999). The secondary structure of proteins is disrupted only at very high pressures, leading to irreversible denaturation, and finally proteins can aggregate in gel (Cheftel, 1995; Hendrickx, Ludikhuyze, Van den Broeck, & Weemaes, 1998; Knorr, 1999; Palou *et al.*, 1999). HHP can also inactivate protease inhibitors such as phytate and increase *in vitro* protein digestibility (IVPD) of legumes (Han, Swanson, & Baik, 2007).

The secondary structure and function of complex polysaccharides and lipids is also modified by HHP (Ledward, 1995). The application of HHP can affect smaller molecules such as vitamin C and β -carotene or inactivate some enzymes (Butz *et al.*, 2002; Bull *et al.*, 2004; Cheftel, 1995).

HHP can also alter the food rheological properties (Patterson *et al.*, 2007). Whereas the physical structure of most high-moisture foods remains unchanged, colour and texture may change after HPP treatment in gas-containing products due to gas displacement and liquid infiltration, leading shape distortion and physical shrinkage, and finally irreversible compression of whole foods (Hogan *et al.*, 2005). However, those modifications can vary in different products; whereas minimal changes in colour, shape and overall appearance can be observed in different fruits such as grapes and blueberries, especially in segments of fruits, similar pressure treatments affected the aspect of green onions and strawberries (Kingsley *et al.*, 2005; O'Reilly *et al.*, 2002). Moreover, changes in colour are minimal in white or cured meats (Cheftel & Culioli, 1997), but colour can be affected in fresh meat and poultry due to modifications in myoglobin, heme displacement/release or ferrous atom oxidation (Hugas, Garriga, & Monfirt, 2002; Cheftel & Culioli, 1997).

HHP inactivation of foodborne viruses in food

Little information on pressure inactivation of viruses exists in comparison to foodborne pathogenic bacteria. Resistance of viruses to HHP depends principally on their structure (Mañas & Pagán, 2005). Whereas non-enveloped viruses show a wide range of sensitivities to HHP (Grove *et al.*, 2006), the viral capsid coat proteins are in general much less stable to HHP than the assembled icosahedral viral particles (Silva, Foguel, Da Poian, & Prevelige, 1996).

The mechanism of virus inactivation by HHP is not well understood. HHP treatment seems not to affect viral nucleic

acids as they can still be detected after the HHP treatment (Khadre & Yousef, 2002; Kingsley, Hoover, Papafragkou, & Richards, 2002; Li *et al.*, 2009; Tang *et al.*, 2010). However, viral nucleic acids can not be detected when an enzymatic treatment is used prior to molecular detection. This enzymatic treatment can eliminate all nucleic acids of naked or disrupted viral particles, and therefore only nucleic acids from intact particles can be detected. It has been suggested that HHP inactivation is produced by the denaturation of the capsid proteins essential for host cell attachment to initiate infection, therefore preventing the binding to host cells (Hogan *et al.*, 2005; Khadre & Yousef, 2002; Kingsley *et al.*, 2002; Buckow & Heinz, 2008; Li *et al.*, 2009; Tang *et al.*, 2010).

The HHP processing parameters, i.e. temperature, time and pressure, can affect the elimination of microorganisms in food products. As a general principle, the degree of inactivation of viruses increase as pressure and/or time increase while the effect of the temperature on HHP virus inactivation varies (Calci, Meade, Tezloff, & Kingsley, 2005; Chen, Hoover, & Kingsley, 2005; Kingsley & Chen, 2008). However, the dissociation and denaturation of proteins and viruses by pressure can be promoted by low temperatures (Bonafe *et al.*, 1998; Foguel, Teschke, Prevelige, & Silva, 1995; Gaspar, Johnson, Silva, & Da Poian, 1997; Kunugi & Tanaka, 2002; Tian, Ruan, Qian, Shao, & Balny, 2000; Weber, 1993; Calci *et al.*, 2005; Chen, Guan, & Hoover, 2006; Kingsley & Chen, 2008). This is due to exposure of non-polar side chains to water at low temperatures. As non-polar interactions are more compressible they are more affected by pressure (Grove *et al.*, 2006; Silva & Weber, 1993).

A non-processing parameter which could also influence the virus inactivation by HHP is the local environment or substrate in which the virus is found (Calci *et al.*, 2005; Chen *et al.*, 2006; Kingsley & Chen, 2008). Food constituents such as proteins, lipids, or carbohydrates can confer a protective effect (Simpson & Gilmour, 1997; García-Graells, Masschalck, & Michiels, 1999; Kingsley & Chen, 2009; Murchie, Kelly, Wiley, Adair, & Patterson, 2007).

Picornaviridae

Viruses from family Picornaviridae are small icosahedral particles which contain a single-stranded positive sense RNA. The family consists of five genera: enteroviruses, rhinoviruses, cardioviruses, aphthoviruses, and hepatoviruses (Lin *et al.*, 2009). Genetic variation among the different genera within this family is considerable. Hepatitis A virus (HAV) (*Hepatovirus* family) is the main foodborne virus of this group, but also poliovirus (*Enterovirus* family) and Aichi virus have been related to food- and waterborne outbreaks (Cliver, 1994; Le Guyader *et al.*, 2008; Choo & Kim, 2006).

The mechanism of inactivation of HAV using HHP treatment has not been clearly unravelled, but results of RNase

protection assays suggest that the HAV capsid remains intact following inactivation by HHP (Kingsley *et al.*, 2002), similarly as observed in rotavirus (Pontes *et al.*, 2001). Therefore, the mechanism of HHP inactivation for HAV is presumed to be denaturation of capsid proteins preventing the attachment to the appropriate cellular receptor, or the blockage of the penetration and virion-uncoating mechanisms subsequent to viral attachment (Kingsley *et al.*, 2002).

Inactivation of HAV has been amply studied in cell culture and food matrices in different conditions, noting considerable differences of HAV inactivation sensitivity in different environments (Table 1). While HHP treatments of HAV stocks in cell culture with pressures of up to 300 MPa had limited effects on HAV titer, higher pressures resulted in significant reduction (Kingsley *et al.*, 2002, 2006; Grove *et al.*, 2008). Treatments with at least 400 MPa are in general the most efficient, but the reductions were significantly different when using different processing temperatures or times (Table 1). The inactivation of HAV is strongly influenced by the temperature. It is reduced proportionally to the decrease of the processing temperature, i.e. it is greater at higher temperatures (>30 °C), while HAV is more resistant to inactivation temperatures close to or below 0 °C (Kingsley, Guan, Hoover, & Chen, 2006) (Table 1). Similarly, the interaction of pH and temperature is also significant. The effect of pH is more evident at 20 °C, and HAV reduction is enhanced throughout the pH range (Kingsley & Chen, 2009).

Oscillatory high-pressure processing -i.e. cycles at high and atmospheric pressure- has been suggested to enhance microbial inactivation (Alemán *et al.*, 1998; Hurtado, Montero, & Borderías, 1998). Interestingly, it does not substantially increase the inactivation rate of HAV (Kingsley *et al.*, 2006). When 2, 4, 6 and 8 cycles were used in treatment with 400 MPa at two different temperatures, 20 and 50 °C, no distinct advantage over continuous high-pressure treatment at the same temperature and pressure conditions was observed (Kingsley *et al.*, 2006).

The salinity of the food environment can confer a protective effect to HHP as salt may act to stabilize viral capsid proteins. While NaCl concentrations up to 1 % does not provide any significant protective effect to HAV, higher concentrations are baroprotective (Kingsley & Chen, 2009; Kingsley *et al.*, 2006; Grove, Lee, Stewart, & Ross, 2009). Similarly, HAV in high salinity seawater required higher pressures for comparable rates of inactivation as observed in isotonic media (Kingsley *et al.*, 2002).

Shellfish is a common source of foodborne viral contamination. A few studies have been carried out on the effect of HHP on HAV in oysters (Table 1). The presence of the shell during commercial bivalve processing does not have any mitigation effects on HHP inactivation as non statistically significant differences are observed when shucked or whole shellfish are treated (Kingsley, Calci, Holliman, Dancho, & Flick, 2009). The salinity of the water where

the oysters are harvested is a key aspect, as intracellular ionic strength of oysters varies with the surrounding water (Kingsley, Holliman, Calci, Chen, & Flick, 2007). Comparable inactivation rates are observed in artificially contaminated shellfish from a low-salinity estuary (approximately 5- to 20-ppt-salinity seawater) to those in normal cell culture at 20 °C using the same pressure and time (Calci *et al.*, 2005). However, the increase of inactivation of HAV in buffers and salts with higher temperatures was not observed for HAV in oyster homogenates (Kingsley & Chen, 2009). This resistance may be due to the composition of oysters that mitigates the inactivation by pressure and high temperature. Nevertheless, there are discrepancies in the results of HAV inactivation in oysters and in buffers with similar pH and NaCl concentration. Whereas Kingsley and Chen (2009) showed that HAV was more resistant in oyster homogenates suggesting that some oyster components are baroprotective, Grove *et al.* (2009) obtained greater HAV inactivation in oyster homogenate than in buffered medium for several pressure and salt combinations (Table 1). These discrepancies may be explained as homogenization of oyster disrupts tissue and membranes and releases cellular contents, being the homogenized tissue therefore exposed to enzymatic degradation. That degradation in combination with HHP can have contributed to the damage of viral capsid proteins which resulted in greater virus inactivation compared to the inactivation observed in buffered medium (Grove *et al.*, 2009).

HAV is less barotolerant in soft fruits and vegetables than in cell culture (Kingsley *et al.*, 2005). In addition, the reduction of HAV in strawberry puree is significantly higher than in sliced green onions using the same conditions (Kingsley *et al.*, 2005). This difference may be related to the acidic pH of strawberry puree (pH 3.67) as the reduction of the pH can increase the HHP inactivation (Patterson *et al.*, 2007; Kingsley & Chen, 2009).

The effect of HHP on HAV attached to pork sausages has been also studied (Sharma *et al.*, 2008). After 5 min treatment of HAV inoculated sausages with 500 MPa at 4 °C, titers recovered from HHP-treated samples were significantly lower. In addition, concomitant chemical (chelating) treatment did not increase virus inactivation on sausages (Sharma *et al.*, 2008).

The susceptibility of the HHP varies among other members of the family Picornaviridae (Table 1). Poliovirus is extremely resistant to HHP, i.e. treatment of 600 MPa for 1 h does not produce significant virus reduction (Wilkinson *et al.*, 2001). Some explanations have been formulated for the poliovirus baroresistance such as the pivotal role of the viral capsid shape (Wilkinson *et al.*, 2001) or the high thermodynamic stability inherent in the composition of the poliovirus particle (Oliveira *et al.*, 1999). Lowering the temperature has no additional effect in poliovirus infectivity. However, the combination of pressure, low temperature (–15 °C) and urea (2 M) produce a significant reduction (Oliveira *et al.*, 1999). This implies a structural change as

Table 1. Effect of HHP treatment on viruses from Picornaviridae family

Virus	Matrix	Pressure	Time	Temp	Reduction ^a	Reference	
Hepatitis A virus	Cell culture	300 MPa	5 min	−10 °C	0.6 PFU/mL	Kingsley et al., 2006	
				40 °C	1.3 PFU/mL		
				50 °C	0.8 PFU/mL		
		350 MPa	1 min	−10 °C	0.9 PFU/mL		
				50 °C	2.4 PFU/mL		
				50 °C	2.8 PFU/mL		
		400 MPa	18 s	−10 °C	0.0 PFU/mL		
				20 °C	0.4 PFU/mL		
			30 s	50 °C	4.1 PFU/mL		
				1 min	−10 °C		1.0 PFU/mL
			10 min	20 °C	2.5 PFU/mL		
				50 °C	4.7 PFU/mL		
		20 min	50 °C	4.9 PFU/mL			
			−10 °C	4.3 PFU/mL			
		Cell culture	300 MPa	10 min	Ambient		>1 TCID ₅₀ /mL
	400 MPa				5 min	1.8 TCID ₅₀ /mL	
	400 MPa				10 min	>2 TCID ₅₀ /mL	
	Cell culture (4.1 ppt NaCl)	450 MPa	5 min		>3.5 TCID ₅₀ /mL (ND)	Kingsley et al., 2002	
				600 MPa	1.5 min		
				450 MPa	15 min		>6 TCID ₅₀ (ND)
	Seawater (27.4 ppt NaCl)	450 MPa	5 min		>3 TCID ₅₀	Kingsley & Chen, 2009	
	Cell culture	400 MPa	1 min	50 °C	4.0 PFU/mL		
	Cell culture with 1 % NaCl				4.1 PFU/mL		
	Cell culture with 3 % NaCl				1.3 PFU/mL	Grove et al., 2009	
	Cell culture with 6 % NaCl				0.4 PFU/mL		
	Cell culture (3 % salt)	300 MPa	10 min	Ambient	<0.5 TCID ₅₀ /mL		
	Oyster homogenate (1.5 % salt)	375 MPa	5 min	Ambient	2 TCID ₅₀ /mL	Kingsley et al., 2002	
5 min					Ambient		1.7 TCID ₅₀ /mL
Oysters	300 MPa	1 min	9 °C	0.2 PFU	Calci et al., 2005		
				0.8 PFU			
				1.3 PFU			
				2.3 PFU			
				3.2 PFU ^b			
Mediterranean mussels	300	5 min	Ambient	0.1 PFU	Terio et al., 2010		
				0.7 PFU			
				1.7 PFU			
				2.5 PFU			
				2.9 PFU			
Blue mussels	300	5 min	Ambient	0.8 PFU	Terio et al., 2010		
				1.0 PFU			
				2.1 PFU			
				2.7 PFU			
				3.6 PFU			
Strawberry puree	250 MPa	5 min	21 °C	1.2 PFU	Kingsley et al., 2005		
				2.1 PFU			
				3.1 PFU			
				4.3 PFU			
Sliced green onions	250 MPa	5 min	21 °C	0.3 PFU	Kingsley et al., 2005		
				0.7 PFU			
				1.4 PFU			
				4.8 PFU			
Sausages immersed in water	500 MPa	5 min	4 °C	3.2 TCID ₅₀ /mL	Sharma et al., 2008		
				2 TCID ₅₀ /mL			
Sausages immersed in 100-ppm EDTA							
Sausages immersed in 2 % lactoferrin					2.1 TCID ₅₀ /mL		

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Table 1 (continued)						
Virus	Matrix	Pressure	Time	Temp	Reduction ^a	Reference
Poliovirus	bisTris-propane buffer (pH 7.5)	240 MPa	120 min	Ambient –15 °C	No significant reduction	Oliveira <i>et al.</i> , 1999
	bisTris-propane buffer (pH 7.5) + 2 M urea	240 MPa	120 min	–15 °C	>4 PFU/mL	
	Cell culture	200 MPa	15 min	20 °C	No significant reduction	Wilkinson <i>et al.</i> , 2001
		400 MPa				
		600 MPa	60 min			
Aichivirus	Cell culture	600 MPa	5 min	Ambient	No reduction	Kingsley <i>et al.</i> , 2002
	Cell culture	600 MPa	5 min	Ambient	<1 TCID ₅₀ /mL	Grove <i>et al.</i> , 2008
	Cell culture	400 MPa	5 min	Ambient	No reduction	Kingsley <i>et al.</i> , 2004
Coxsackievirus B5	Cell culture	500 MPa	5 min	Ambient	No reduction	
		600 MPa				
		400 MPa				
Coxsackievirus A9	Cell culture	400 MPa	5 min	Ambient	3.4 TCID ₅₀	
		500 MPa			6.5 TCID ₅₀	
		600 MPa			>7.1 TCID ₅₀ (ND)	

ND: non-detected.
^a Results shown are log₁₀ reductions observed or calculated from the text and tables of references.
^b Virus was not detected in one of three trials, so the detection limit 1.5 log₁₀ was assumed.

urea predominantly affects the secondary and tertiary structure of proteins, primarily by perturbing the hydrogen-bonding network.

Coxsackievirus and Aichi virus display variable sensitivities to HHP (Table 1). While coxsackievirus A9 can be significantly reduced with moderate treatments, coxsackievirus B5 and Aichi virus were completely resistant to HHP treatments at 600 MPa during 5 min (Kingsley, Chen, & Hoover, 2004).

Caliciviridae

The Caliciviridae family includes two genera that infect humans: noroviruses and sapoviruses. Noroviruses are the most common cause of outbreaks and sporadic cases of acute gastroenteritis (Noda, Fukuda, & Nishio, 2008). They possess a single-stranded, positive sense RNA genome, surrounded by an icosahedral capsid (Greening, 2006). Viruses from the *Norovirus* genus were recently classified in 29 genetic clusters within five genogroups (Zheng *et al.*, 2006) from which most of the human noroviruses belong to the genogroups I and II (Patel, Hall, Vinjé, & Parashar, 2009). Outbreaks caused by viruses from these groups are often a primary result of exposure to contaminated food or water, while further propagation is normally spread among contacts with primary cases (Becker, Moe, Southwick, & MacCormack, 2000). As human noroviruses have not been reliably propagated in cell cultures and there are no suitable animal models for their propagation, two surrogate viruses belonging to the family Caliciviridae, feline calicivirus (FCV) and more recently murine norovirus (MNV), are normally used in inactivation studies (Doultree, Druce, Birch, Bowden, & Marshall, 1999; Hewitt, Rivera-

Aban, & Greening, 2009) (Table 2). However, the interpretation of the results of inactivation studies using surrogates must be carefully considered, as for example MNV is more resistant to HHP than FCV; e.g. total FCV inactivation and only 1.8 PFU/mL MNV reduction with ~ 300 MPa for 5 min (Kingsley *et al.*, 2002, 2007; Murchie *et al.*, 2007).

FCV inactivation in cell culture by HHP increases in parallel with the increase of pressure and/or time of application, and low combinations are enough to inactivate FCV to undetectable levels (Chen *et al.*, 2005; Kingsley *et al.*, 2002; Grove *et al.*, 2008) (Table 2). However, the pressure levels have a more dramatic effect on virus inactivation than the operational time, and the same level of inactivation can be obtained with a minimum increase of pressure and a severe reduction of treatment time, e.g. an increase of just 50 MPa and 90.8 % reduction of operational time (Chen *et al.*, 2005).

The effect of temperature has been also studied in the FCV inactivation by HHP. The inactivation rate at temperatures close to ambient is acceptable, however higher (above 50 °C) and lower (below 0 °C) temperatures are more effective (Chen *et al.*, 2005; Buckow, Isbarn, Knorr, Heinz, & Lehmacher, 2008) (Table 2). Interestingly, the effects of the temperature on HHP inactivation of FCV differs from those observed in HAV, where the inactivation only increases at temperatures above 30 °C (Kingsley *et al.*, 2006). Another interesting finding is that the inactivation of FCV is more efficiently inactivated in mineral water than in cell culture (Buckow *et al.*, 2008).

Higher acidity, ionic concentration (NaCl), and sucrose concentration, which can be found in some foods, can have a substantial effect on inactivation of FCV by HHP.

Table 2. Effect of HHP on viruses from Calciviridae family

Virus	Matrix	Pressure	Time	Temp.	Reduction ^a	Reference	
Feline calicivirus	Cell culture	200 MPa	5 min	Ambient	3.4 TCID ₅₀	Kingsley et al., 2002	
		225 MPa			4 TCID ₅₀		
		250 MPa			5.7 TCID ₅₀		
		275 MPa			>6.6 TCID ₅₀ (ND)		
		300 MPa			>6.6 TCID ₅₀ (ND)		
	Cell culture	200 MPa	4.5 min	ambient	1 TCID ₅₀	Grove et al., 2008	
		265 MPa	3 min		>5 TCID ₅₀		
		300 MPa	2 min		3.6 TCID ₅₀		
		300 MPa	3 min		>5 TCID ₅₀ (ND)		
		450 MPa	0.5 min		>5 TCID ₅₀ (ND)		
	Cell culture	200 MPa	4 min	–10 °C	5 PFU/mL	Chen et al., 2005	
					0 °C		4.4 PFU/mL
					20 °C		0.3 PFU/mL
					50 °C		4 PFU/mL
					20 min		21 °C
		76 min	21 °C	3.7 PFU/mL			
		250 MPa	7 min	21 °C	3.9 PFU/mL	Buckow et al., 2008	
		200 MPa	6 min	5 °C	3 PFU		
		250 MPa	4 min	5 °C	5 PFU		
		300 MPa	0.5 min	10 °C	2 PFU		
	450 MPa	2 min	75 °C	>7 PFU/mL			
	Cell culture	150 MPa	5 min	20 °C	0.3 TCID ₅₀	Murchie et al., 2007	
					200 MPa		1.3 TCID ₅₀
					250 MPa		3.8 TCID ₅₀
					300 MPa		>5.4 TCID ₅₀ (ND)
					350 MPa		>5.4 TCID ₅₀ (ND)
	Cell culture pH 6	250 MPa	1 min	20 °C	4.1 PFU/mL	Kingsley & Chen, 2008	
	Cell culture				5 min		20 °C
	Cell culture + 12 % NaCl				0.7 PFU/mL		
	Cell culture + 6 % NaCl				1.9 PFU/mL		
	Cell culture + 20 % sucrose				4.0 PFU/mL		
	Cell culture + 40 % sucrose				0.9 PFU/mL		
	Cell culture + 6 % NaCl and 20 % sucrose				0.9 PFU/mL		
	Cell culture	200 MPa	5 min	4 °C	4.7 PFU/mL	Buckow et al., 2008	
					Cell culture + 6 % NaCl		2.7 PFU/mL
					Cell culture + 20 % sucrose		3.3 PFU/mL
					Cell culture + 6 % NaCl and 20 % sucrose		1.7 PFU/mL
	Mineral water	200 MPa	6 min	5 °C	5 PFU	Buckow et al., 2008	
		300 MPa	0.5 min	10 °C	6 PFU		
		450 MPa	1 min	15 °C	>7 PFU/mL		
	Seawater	150 MPa	5 min	20 °C	–0.3 TCID ₅₀	Murchie et al., 2007	
					200 MPa		0.5 TCID ₅₀
250 MPa					3.5 TCID ₅₀		
300 MPa					>4.1 TCID ₅₀ (ND)		
350 MPa					>4.1 TCID ₅₀ (ND)		
Mussels	150 MPa	5 min	20 °C	0.6 TCID ₅₀	Murchie et al., 2007		
				200 MPa		1.0 TCID ₅₀	
				250 MPa		1.4 TCID ₅₀	
				300 MPa		>3.2 TCID ₅₀	
				350 MPa		>4.2 TCID ₅₀ (ND)	
Oysters	150 MPa	5 min	20 °C	0.1 TCID ₅₀	Murchie et al., 2007		
				200 MPa		0.6 TCID ₅₀	
				250 MPa		1.6 TCID ₅₀	
				300 MPa		>3.8 TCID ₅₀ (ND)	
				350 MPa		>3.8 TCID ₅₀ (ND)	
Sausages immersed in water	500 MPa	5 min	4 °C	2.9 TCID ₅₀ /mL	Sharma et al., 2008		
				Sausages immersed in 100-ppm EDTA		2.4 TCID ₅₀ /mL	
				Sausages immersed in 2 % lactoferrin		2 TCID ₅₀ /mL	

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Virus	Matrix	Pressure	Time	Temp.	Reduction ^a	Reference		
Murine norovirus	Cell culture	350 MPa	5 min	5 °C	5.6 PFU/mL	Kingsley <i>et al.</i> , 2007		
				10 °C	4.8 PFU/mL			
				20 °C	1.8 PFU/mL			
				30 °C	1.2 PFU/mL			
	Oyster tissue	400 MPa	5 min	5 min	20 °C	6.9 PFU/mL	Tang <i>et al.</i> , 2010 Li <i>et al.</i> , 2009	
					400 MPa	0 °C		8.2 PFU
					200 MPa	0 °C		0.5 PFU
					300 MPa			0.9 PFU
Oyster tissue	400 MPa	5 min	5 min	400 MPa	>4.1 PFU (ND)	Kingsley <i>et al.</i> , 2007		
				400 MPa	5 °C		4.1 PFU/mL	

ND: non-detected.
^a Results shown are log₁₀ reductions observed or calculated from the text and tables of references.

A low pH environment *per se* results in a substantial reduction of FCV in non-pressurized samples, but there is almost no additional reduction at pH ≤ 5.2 when samples are pressured (Kingsley & Chen, 2008). Conversely, substantial reduction occurs when FCV is pressured at pH ≥ 6, with the highest reduction at pH 6 (Kingsley & Chen, 2008) (Table 2). As was observed for HAV, the effect of NaCl on FCV is baroprotective. The increasing protection effect is evident from 0 to 12 % NaCl, but no significant increase of the protection is observed when samples are supplemented with concentrations higher than 12 % or up to 21 % (Table 2). Similar effects are observed for sucrose. An enhanced protection effect is observed when the sucrose concentration is increased up to 40 % (Table 2). However, higher sucrose concentrations (i.e. up to 70 %) do not significantly increase the baroprotective effect. Interestingly, when both sucrose and NaCl are added, the baroprotective effect on FCV is additive (Table 2). A reduction in water activity generally results in greater pressure resistance, however, similar water activities for NaCl and sucrose solution result in different levels of FCV baroprotection meaning the degree of pressure inactivation of FCV was not simply a function of water activity.

The HHP effect on FCV has been shown to be different in the different foods studied. Whereas no significant viral reduction was observed regardless of the matrix studied with a mild HHP treatment (i.e. 150 MPa for 5 min at 20 °C), a moderate increase in pressure (up to 250 MPa) produced an obvious difference in inactivation, which was lower in mussels and oysters in comparison to seawater or cell culture (Murchie *et al.*, 2007) (Table 2). These results are in agreement with those obtained using HAV, which suggests that some components in oysters can be baroprotective (Kingsley & Chen, 2009). However, 300 MPa was enough for total reduction of virus in all matrices, but it should be considered that the initial viral concentrations were not the same in the matrices (Table 2). The effect of HHP on FCV attached to pork sausages has been also studied (Sharma *et al.*, 2008). As for HAV, a significant reduction was observed after 5 min treatment with 500 MPa at

4 °C without any additional effect of a concomitant chemical (chelating) treatment (Sharma *et al.*, 2008).

Another human norovirus surrogate has been used in the latest years: murine norovirus (MNV). The results of the inactivation studies using MNV may be more relevant for human norovirus as the two viruses share biochemical and molecular similarities, and an identical route of infection (Wobus, Thackray, & Virgin, 2006). The effect of processing parameters (operational pressure, time and temperature) on MNV inactivation is similar to other viruses (Table 2): significant reductions using increasing pressures (from 325 MPa to 450 MPa) or times or at low temperatures (Kingsley *et al.*, 2007). The inactivation increases in parallel to the increase of operational pressure and/or time, with a rapid initial reduction followed by tailing at longer treatment times as also observed for other viruses.

The effect of HHP inactivation on MNV has been also studied in foods, such as oysters (Kingsley *et al.*, 2007; Li *et al.*, 2009) (Table 2). A slight MNV reduction was obtained in oysters treated with mild pressures (200 and 300 MPa), but a significant reduction (above 4 log reduction) is observed at 400 MPa (Kingsley *et al.*, 2007; Li *et al.*, 2009) (Table 2).

Binding of MNV to RAW 264.7 cells declined remarkably after HHP treatment indicating that the attachment of MNV is affected by HHP. Those results show that HHP primarily affects the receptor-binding site of the MNV capsid protein, suggesting a possible means for development of a NoV vaccine that contains virus inactivated with HHP treatment, if human NoV has similar susceptibilities to HHP as MNV (Tang *et al.*, 2010).

Reoviridae

Rotaviruses are members of Reoviridae family, and are non-enveloped viruses with icosahedral capsid of 60–80 nm in diameter and possess linear segmented double stranded RNA genome. They are involved in acute food and waterborne gastroenteritis, especially in children (Greening, 2006). Khadre and Yousef (2002) used a come-up time strategy for HHP inactivation of rotavirus,

Table 3. Effect of HHP on viruses from Reoviridae family

Virus	Matrix	Pressure	Time	Temp	Reduction ^a	Reference
Rotavirus	Cell culture	0–300 MPa	1.2 min ^b	25 °C	5 TCID ₅₀ /mL	Khadre & Yousef, 2002
			2 min		8 TCID ₅₀ /mL	
			4 min		9 TCID ₅₀ /mL	
		300 MPa	6 min		8 TCID ₅₀ /mL	
			8 min		9 TCID ₅₀ /mL	
			10 min		9 TCID ₅₀ /mL	
		500 MPa	10 min		9 TCID ₅₀ /mL	
			800 MPa		10 min	

^a Results shown are log₁₀ reductions observed or calculated from the text and tables of the referenced article.
^b Pressure come-up time.

and 5 log TCID₅₀/mL reduction was observed when 300 MPa at 25 °C during the come-up time, 70 s, was used (Table 3). With 300, 500 and 800 MPa treatments for 2–10 min at 25 °C, 8 or 9 log TCID₅₀/mL reduction was observed (Table 3). Interestingly, times longer than 2 min did not provide any additional decrease of rotavirus titer.

Conclusions

HHP is a promising processing technique for food industry as it offers numerous opportunities for developing novel applications. Besides efficient disinfection, a selection of minimally processed safe foods can be foreseen. HHP produced food items are safe for consumers with extended shelf life, high nutritional value and excellent sensorial characteristics (Fonberg-Broczek *et al.*, 1999, 2005; Buckow & Heinz, 2008). Besides HHP applications in food safety area, other applications are indicated. HHP is currently used, for example, for shucking of oysters or to facilitate the removal of the shell of crustacean shellfish such as lobster, crab and shrimp (Terio *et al.*, 2010). HHP has also been successfully evaluated as a potential method for preparation of vaccines and it can be used in some industrial processes such as modulation of microbial fermentations, or it can influence biosynthesis pathways and thus lead to the formation of product variants with novel functional properties (Aertsen, Meersman, Hendrickx, Vogel, & Michiels, 2009).

Although it is accepted that HHP can inactivate foodborne viruses, there are some important technological aspects that must be considered. The balance between food safety and food quality must be considered for each particular food and virus. It is necessary to apply the correct pressure conditions that efficiently eliminate viruses (and other pathogens) without affecting the food quality. In addition short treatment times are desired for economical and nutritional reasons. HHP conditions must be determined independently for each type of virus as the response and susceptibility is heterogeneous, i.e. from a severe (e.g. hepatitis A or murine norovirus) to a slight reduction (e.g. poliovirus). Identical HHP conditions do not produce

similar reductions in members of the same virus family. Therefore, the definition of standard HHP processing criteria, i.e. selection of horizontal processing parameters such as combination of pressure, temperature and time, must be taken carefully in order to assure safe products for final consumers.

Another important aspect in disinfection studies is that some of the main enteric viruses can not grow in cell culture, so there is a lack of direct evaluation of inactivation. In order to overcome these problems, the use of non-pathogenic virus surrogates of similar structural characteristics has been suggested. Mengo virus MC₀ (Costafreda, Bosch, & Pintó, 2006) and feline calicivirus and murine NoV-1 (Cannon *et al.*, 2006) have been proposed as ideal surrogates for HAV and human NoV, respectively. However, it is still under question if the results generated using viral surrogates can be precisely extrapolated to the target viruses.

Responses of foodborne viruses to HHP can vary; however they follow similar kinetic models of inactivation. The development of exact mathematical models for prediction of HHP virus inactivation can be beneficial for the food industry, as they would be useful for optimizing process conditions and constructing hazard analysis critical control point programs to guarantee food safety (Chen *et al.*, 2005). Different models have been used to predict the HHP inactivation of viruses, such as linear or non-linear (Weibull or log-logistic) models. They differ in the assumption that the cells in a population have the same (linear model) or different (Weibull model) resistance to lethal treatments. The non-linear models have been recognized as more appropriate to describe pressure inactivation of viruses (Kingsley *et al.*, 2006, 2007; Grove *et al.*, 2009; Chen *et al.*, 2005). However, poor functional relationships with pressure and temperature in the secondary model approach have been also noticed (Buckow *et al.*, 2008).

In conclusion, although HHP is shown to be a promising strategy for inactivation of microorganisms, it is still in an initial stage for foodborne viruses. Important aspects must be clearly addressed such as the reasons underpinning the differences in resistance of foodborne viruses to HHP in different food products, or the definition of new strategies for

the evaluation of the inactivation results for viruses that can not grow in cell culture. Consequently, more inactivation studies using a range of processing and technological parameters for different food products and viruses are needed to clearly determine the conditions for efficient removal of foodborne viruses. This will also contribute to elucidate the mechanisms of HHP inactivation on viruses. Finally, it will help to define and develop predictive inactivation models for practical application in modern food processing.

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